

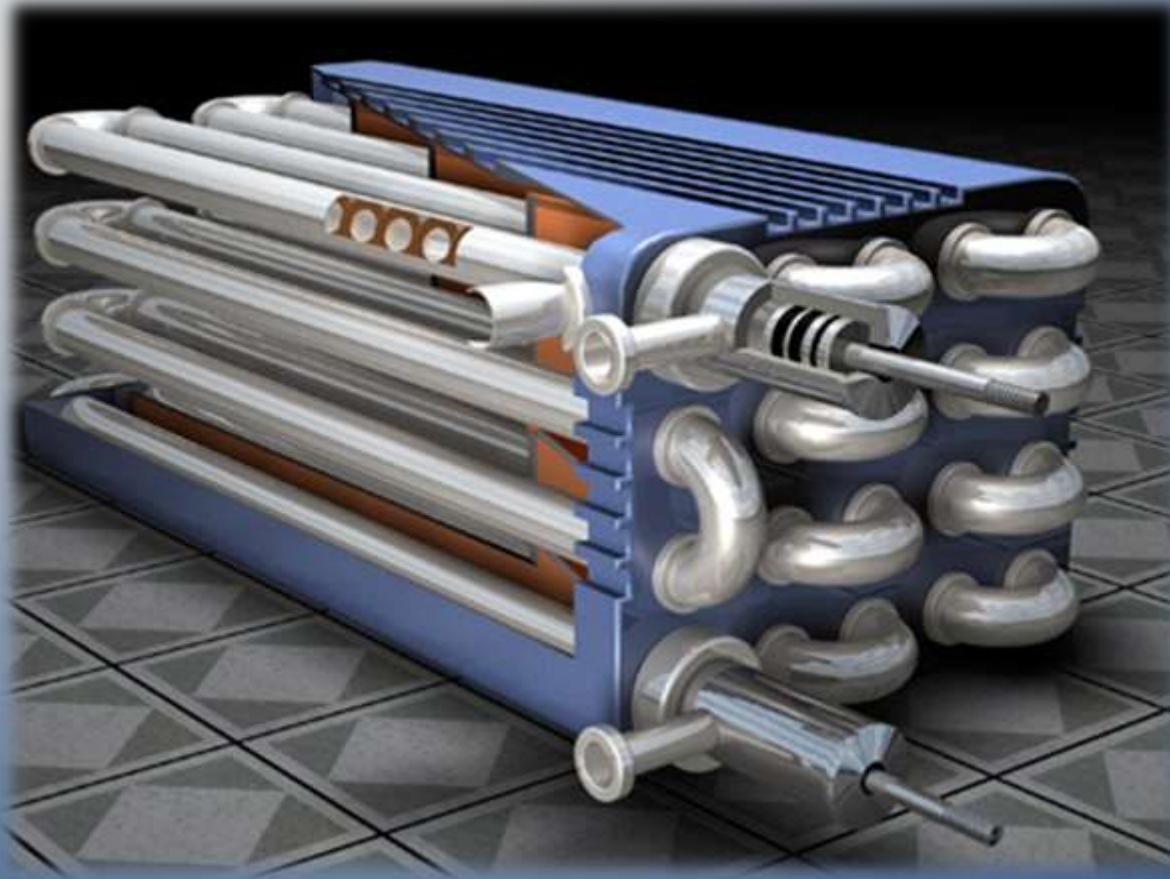
University of KwaZulu Natal  
Chemical Engineering

A New Group Contribution  
Method For The Estimation Of  
Thermal Conductivity For Non-  
Electrolyte Organic Compounds

Onellan Govender

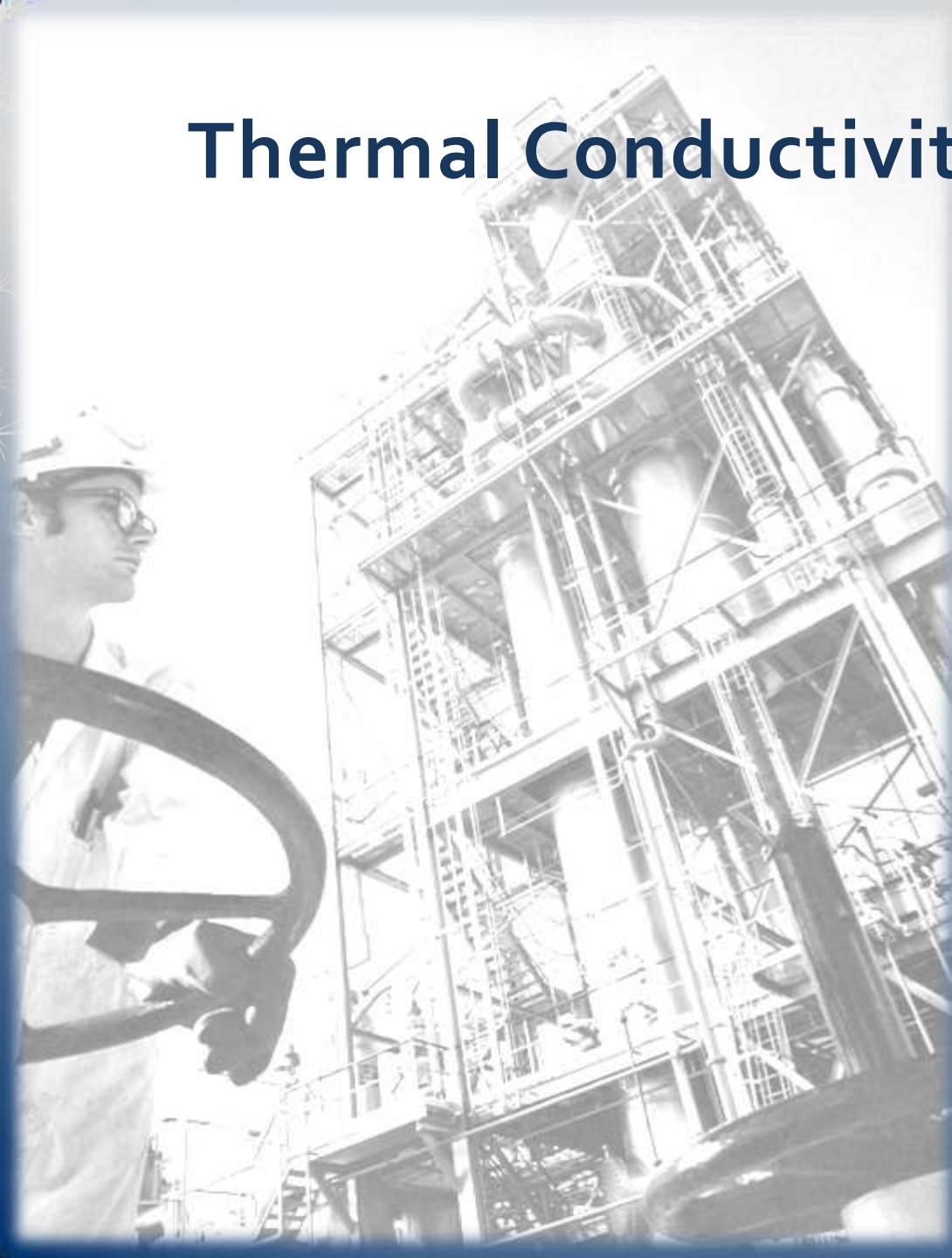
205502080

# Experimentation Verses Prediction



# Thermal Conductivity Required for

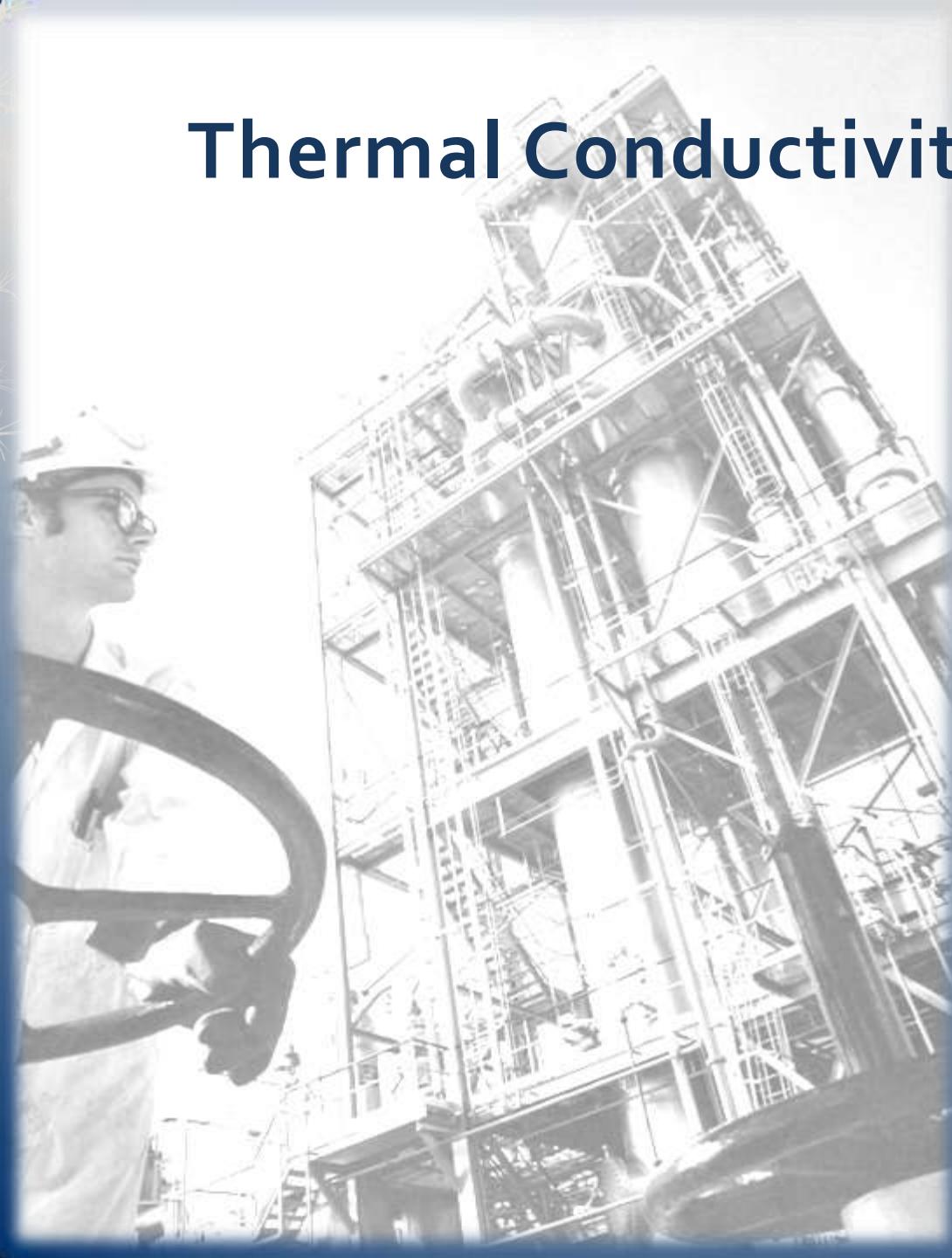
- ★ Equipment design
- ★ Cost-effective and safe plant design
- ★ Simulation packages
- ★ Scale up
- ★ Calculation of transfer coefficients and dimensionless numbers





# Thermal Conductivity Required for

- ★ Equipment design
- ★ Cost-effective and safe plant design
- ★ Simulation packages
- ★ Scale up
- ★ Calculation of transfer coefficients and dimensionless numbers



# Previous Work Done

- ✿ Normal boiling point (Rarey and Cordes 2002 and Nannoolal *et al.* 2004)
- ✿ Critical property data (Nannoolal *et al.* 2007)
- ✿ Vapour pressures (Nannoolal *et al.* 2008 and Moller *et al.* 2008)
- ✿ Liquid viscosity (Nannoolal *et al.* 2009)

# Current Work

- ✿ Bruce Moller – Gamma infinity in water, alkanes, alcohols
- ✿ Eugene Olivier – Surface Tension
- ✿ Onellan Govender – Thermal Conductivity

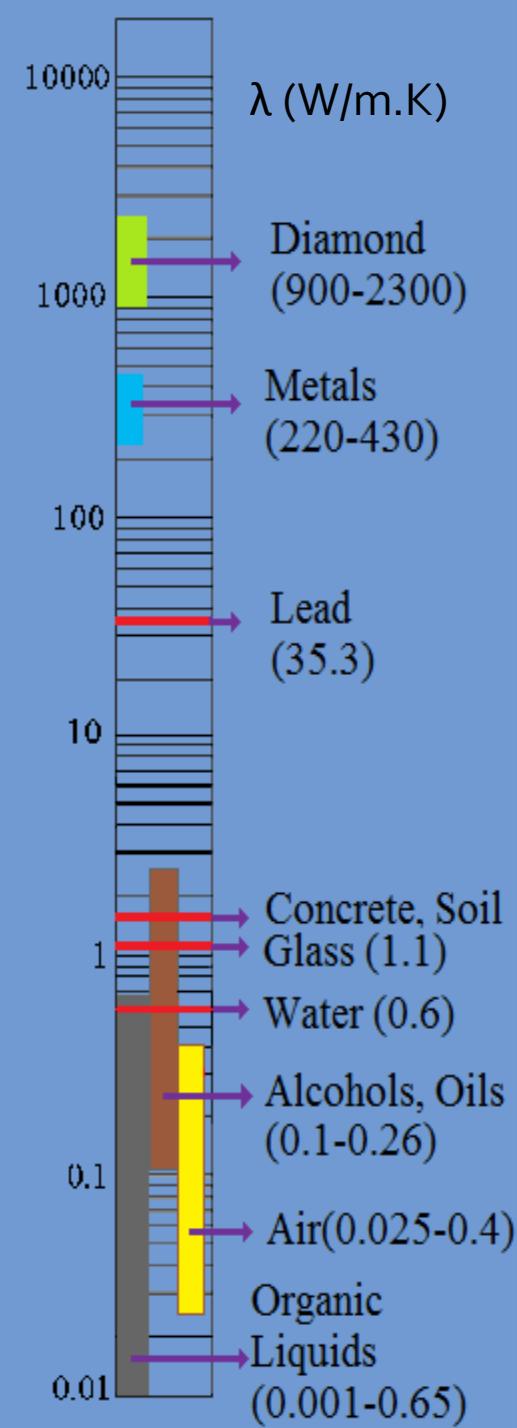
# Thermal Conductivity - $\lambda$

$$J_q = -\lambda \nabla T$$

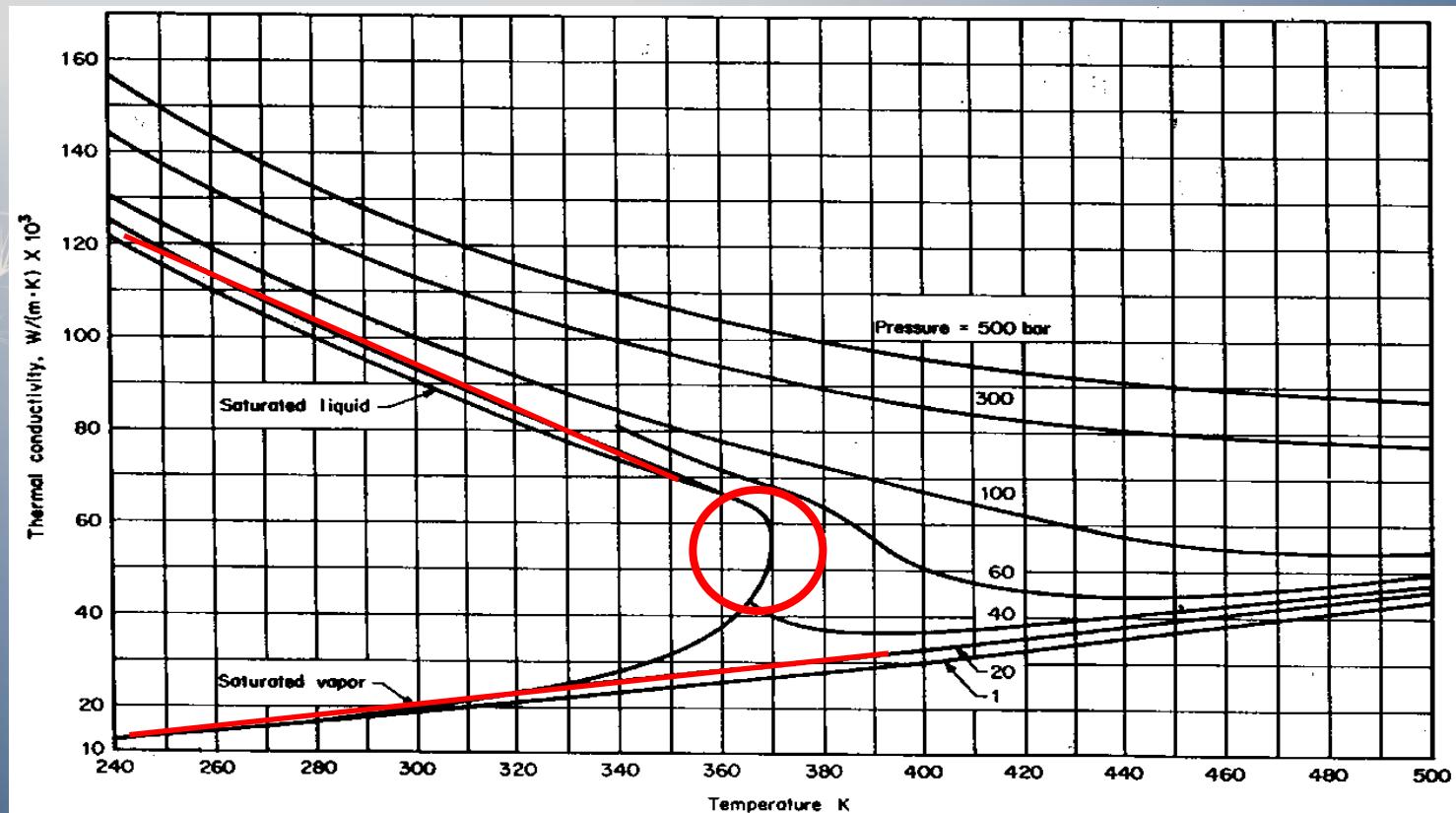
- \* Theoretical Contributions and Considerations

- \* Critical Enhancement

$$\lambda(\rho_n, T) = \lambda_o(T) + \Delta\lambda(\rho_n, T) + \Delta\lambda_c(\rho_n, T)$$



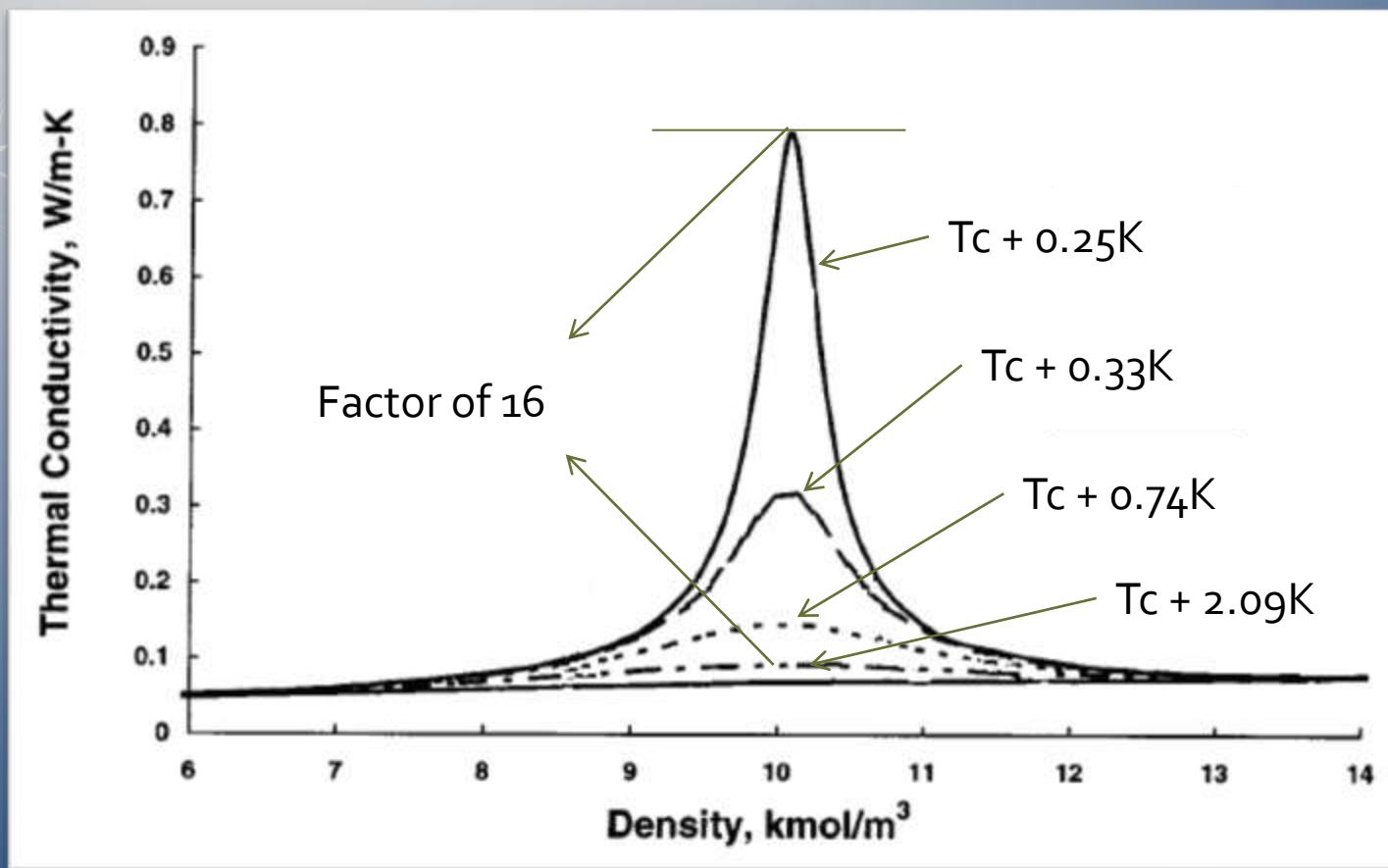
# Thermal Conductivity of Propane as f(T,P)



$$\lambda(\rho_n, T) = \lambda_o(T) + \Delta\lambda(\rho_n, T) + \Delta\lambda_c(\rho_n, T)$$

# Critical Enhancement

Selected isotherms for CO<sub>2</sub> depicting the critical enhancement phenomenon (Mathias *et al.* 2002)



# Correlation and Prediction

## ✿ Experiments and Data Correlation

## ✿ Prediction Models

- ✿ Corresponding States

- ✿ General Correlations

- ✿ Family Methods

- ✿ Group Contribution Method

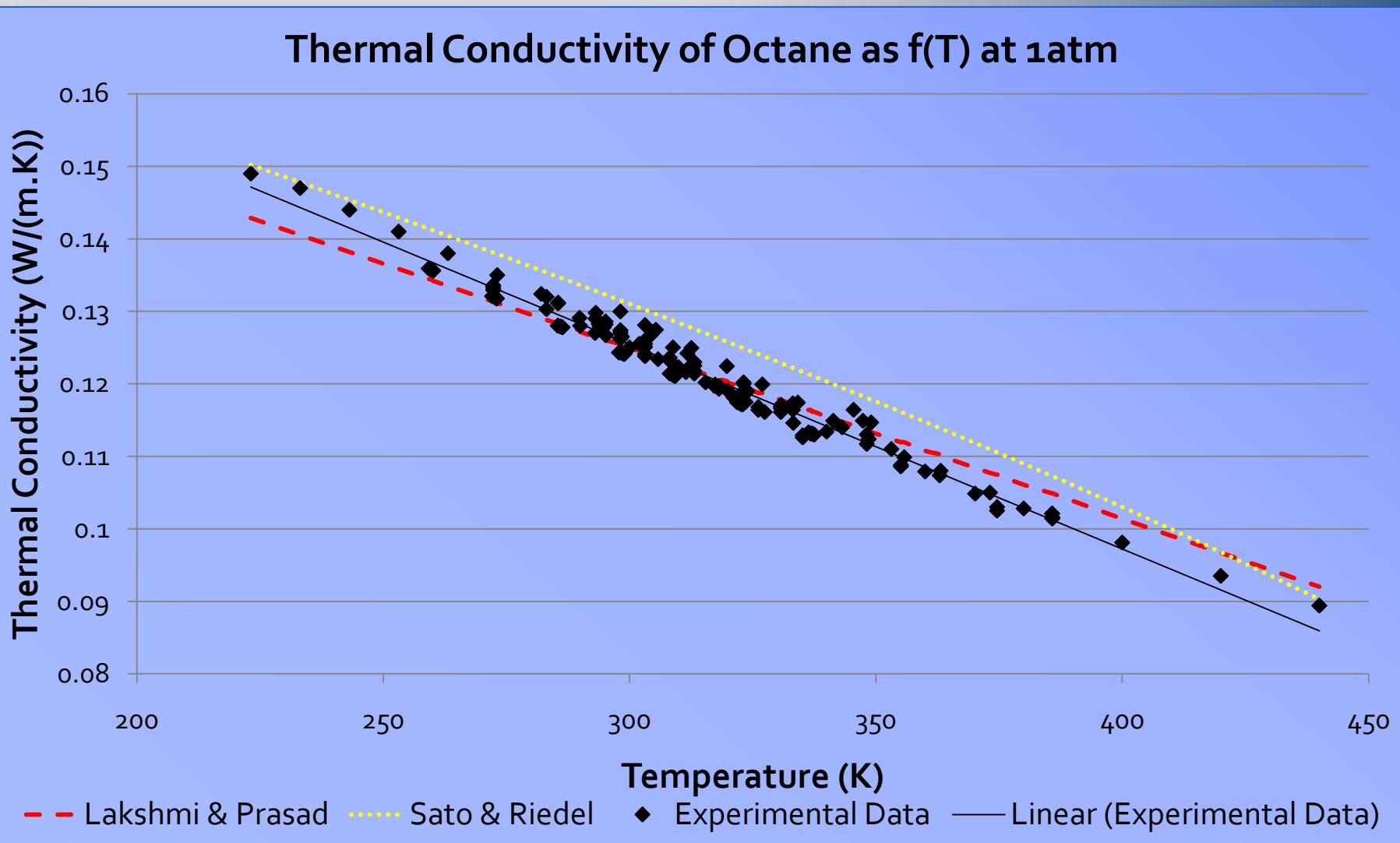
# General Correlation Methods

- ✿ Sato and Reidel (1977)
- ✿ Lakshmi and Prasad (1992)

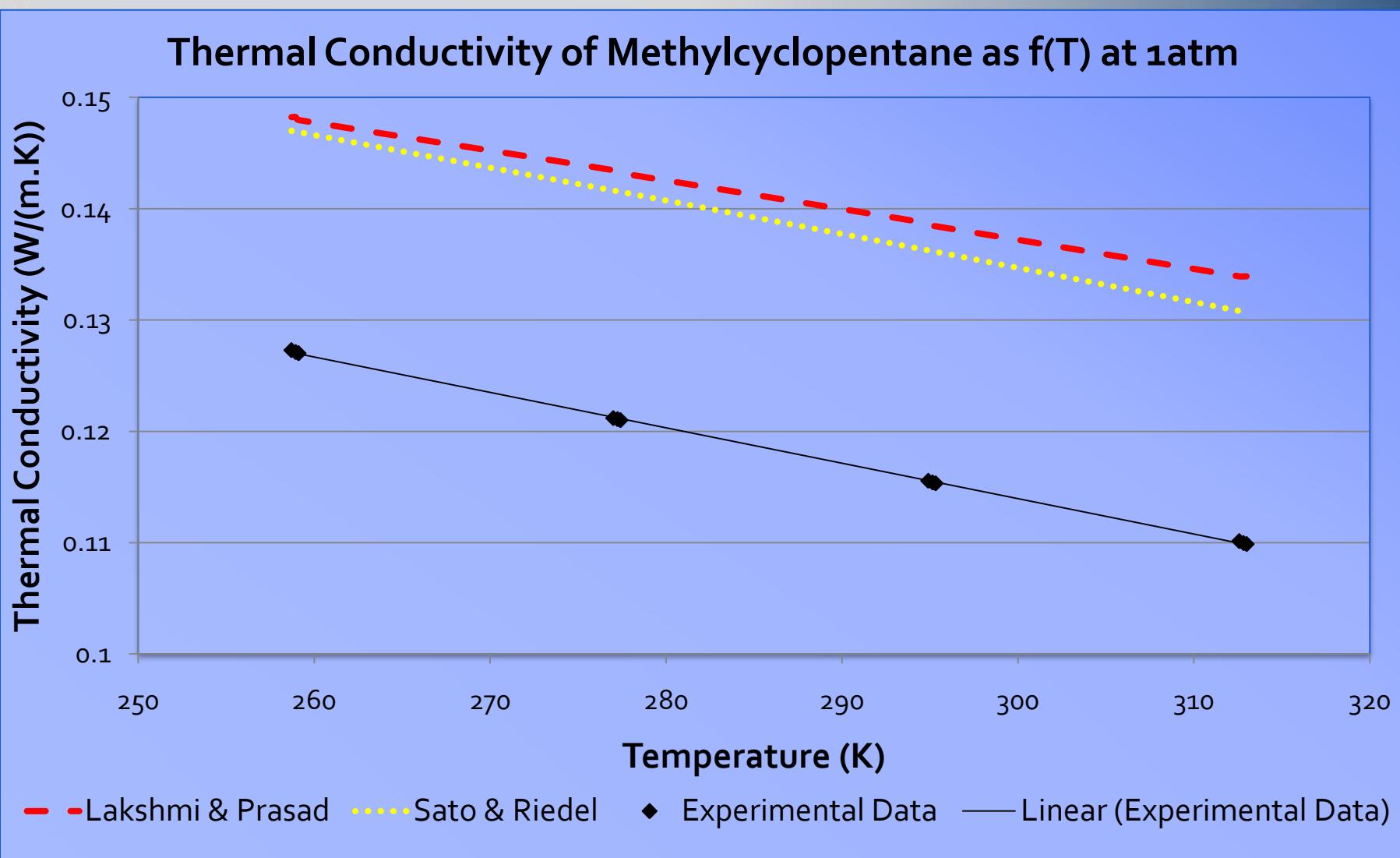
## Family Methods

- ✿ Latini *et al.* (1977)

# General Correlation Methods



# General Correlation Methods



# Previous Group Contribution Methods

- ✿ Sakiadis and Coates (1955 , 1957)
- ✿ Robbins & Kingrea (1962)
- ✿ Nagvekar & Daubert (1987)
- ✿ Assael, Charitidou & Wakeham (1989)
- ✿ Sastri and Rao (1993)
- ✿ Rodenbush, Viswanath & Hsieh (1999)
- ✿ Sastri and Rao (1999)

# Literature Review & Method Test

Component Classes			
Hydrocarbons	Oxygen Compounds	Ethers	Aldehydes
Aromatic Hydrocarbons	Carboxylic Acids	Esters	Nitrogen Compounds
Halogen Compounds	Alcohols	Ketones	Amines

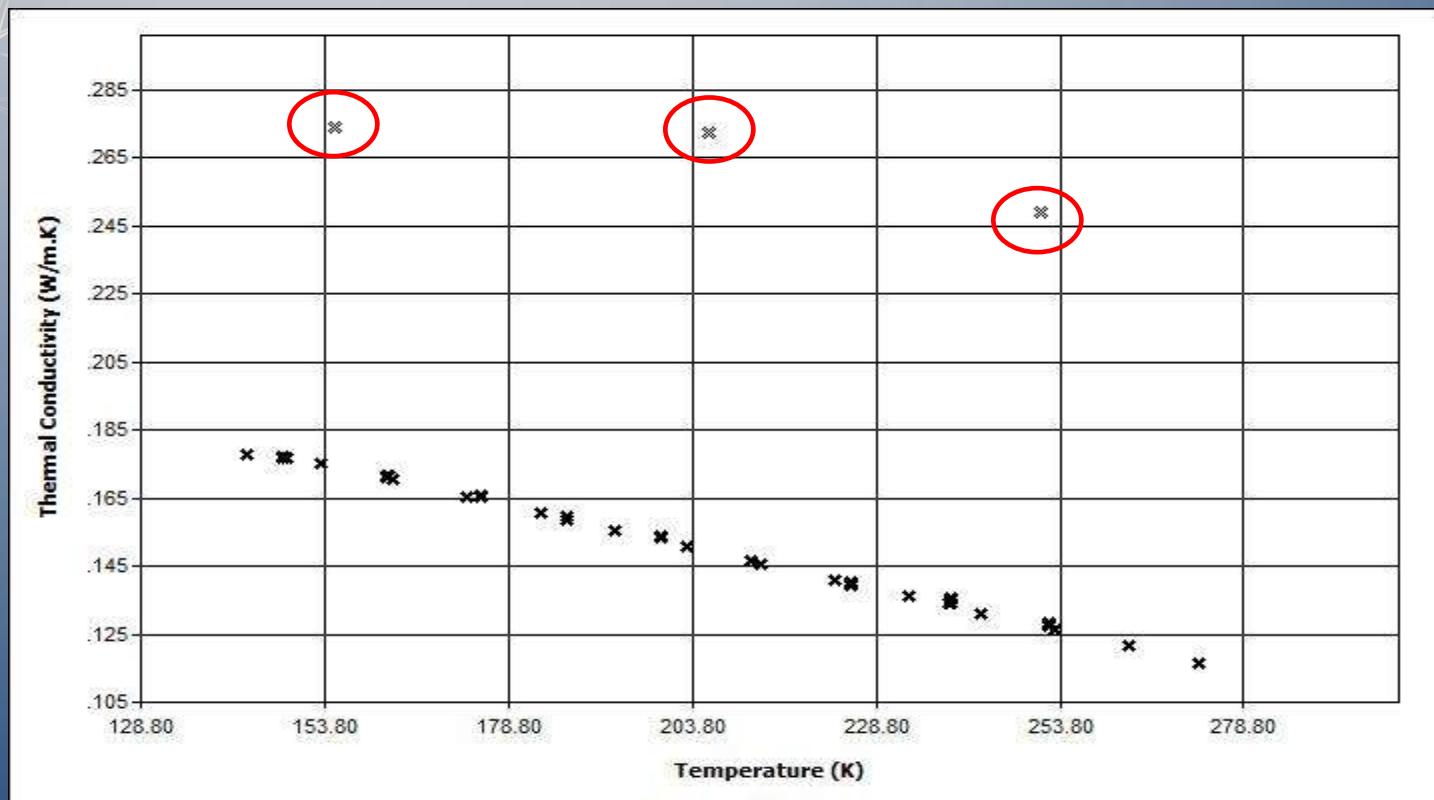
Methods	Sato & Riedel (1997)	Nagvekar & Daubert (1987)	Lakshmi & Prasad (1992)	Sastri & Rao (1993)
RMD (%)	19.81	16.64	23.41	14.11
Number of Components	500	322	500	469

# Data Filtration and Validation

DDB – 100 515 data points for 876 components

Figure : Data points for n-butane

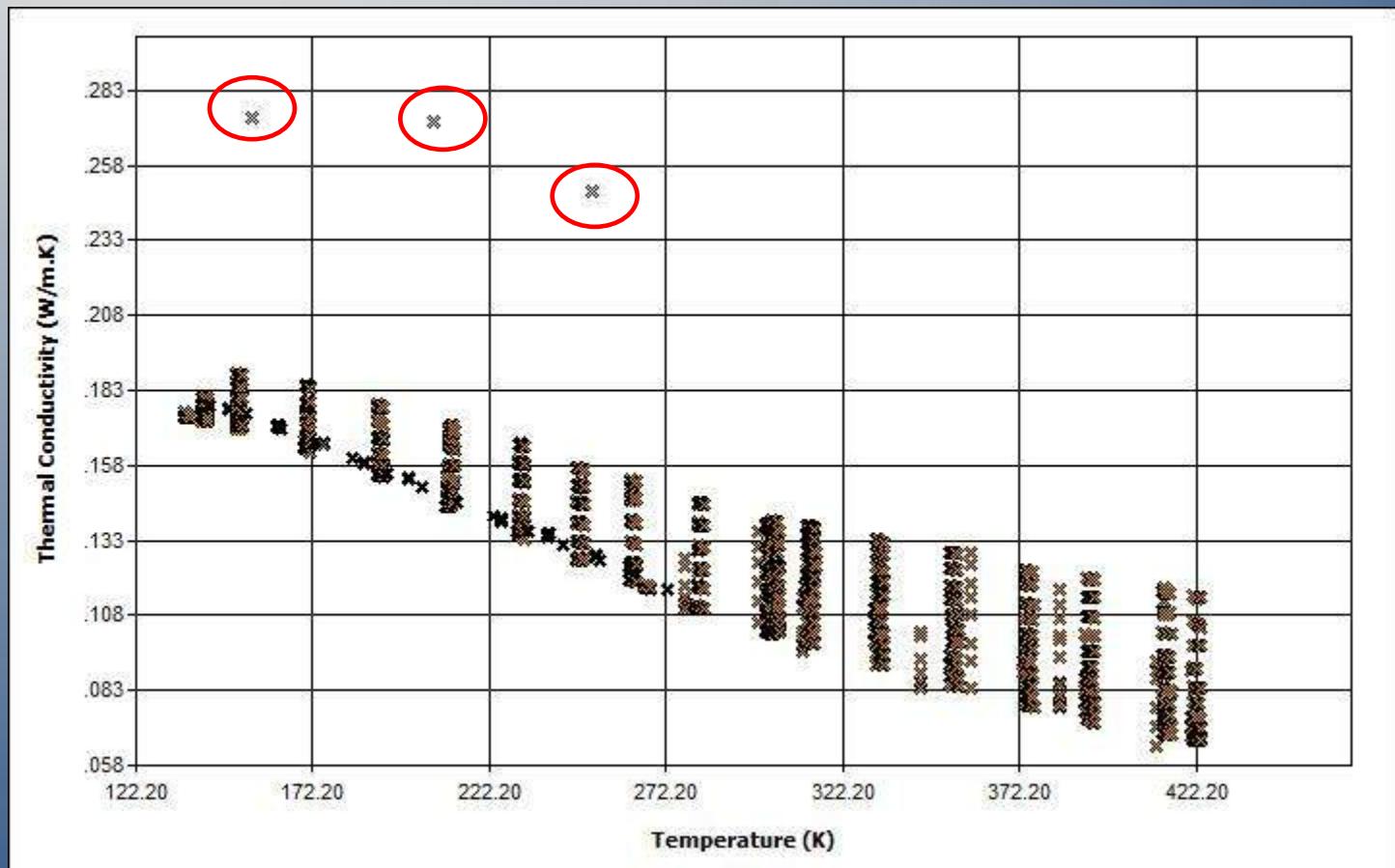
(1atm isobar;  $135.75 < T \text{ (K)} < 272.65$ )



# Data Filtration and Validation

Figure : All experimental data points for n-butane

( $135.75 < T \text{ (K)} < 423.61$ ;  $101.325 < P \text{ (kPa)} < 70000$ )



# Model Idea

Separate group contributions for the 3 parts

$$\lambda(T, P) = \lambda(T_{Ref}) \times f(T - T_{Ref}) \times g(P - 101.3kPa)$$

↑                           ↑                           ↑

Reference Value      Temperature Dependence      Pressure Dependence

$$\lambda = \lambda_{REF} + (T - T_{REF}) \times \frac{\partial \lambda}{\partial T} + \lambda_P$$

↓                           ↓                           ↓

$$\lambda_P = (P - 101.325) \times \left[ \frac{\partial \lambda}{\partial P} + \left( [T - T_{REF}] \times \frac{\partial \lambda / \partial P}{\partial T} \right) \right]$$

# Software Development



**Data Validation**

Load/Save  
Data Source: Random  
Save Changes | Exit  
Flag Component as Done

Current Verification  
DDB number:   
Prev DDB #: Next DDB #:  
Organic Group: All  
Go to DDB Number: 99  
Clear Graph | Plot Graph  
Plot Non-Pres Dep. Sing. Points

P (kPa)	A (mJ/m)	T (K)	Flag
101.325	0.1598	183.15	0
101.325	0.1568	193.15	0
9000	0.152	196.65	?
30000	0.1598	196.65	?
40000	0.161	196.65	?
101.325	0.1505	196.65	1
50000	0.1633	196.65	?
100000	0.1533	196.65	?
20000	0.156	196.65	?
600	0.1513	203.15	?
10000	0.153	203.15	?
20000	0.1579	203.15	?
30000	0.1608	203.15	?
40000	0.1639	203.15	?
101.325	0.1533	203.15	0
10000	0.1522	213.15	?
600	0.1483	213.15	?
30000	0.1582	213.15	?
40000	0.1612	213.15	?
101.325	0.1568	213.15	0
20000	0.1553	213.15	?
9000	0.149	220.65	?
50000	0.1582	220.65	?
101.325	0.1433	220.65	1
40000	0.1557	220.65	?
10000	0.1467	220.65	?
30000	0.1528	220.65	?
20000	0.1499	220.65	?

Thermal Conductivity (W/mK)

Temperature (K)

Data Flagging  
Flag Point | Remove Flags  
Flag Pressure | Flag Duplicates  
Thermal Cond. Tol.: 0.0001  
Temperature Tol.: 0.05

Plot Saito & Rao for current DDB number.  
Plot Lajoski for current DDB number.  
Plot Neguski & Daubert for current DDB number.

Critical Temp (K): 507.4  
Boiling Point (K): 343.1  
Export Graph as jpeg  
Export S Graph as jpeg  
Grit Press (kPa): 2975

Linear Regression of Data  
Linear Regression  
TC Ref: m.TC/P  
TREF: m.TCP/T  
m.TC/T AAD:

**Data Regression**

EXIT | Save Changes | Flag All 1

Organic Group: All  
DDB Number: 99  
Hexane  
C6H14  
NEXT

plot flag = 0 | Plot TCN Vs T | Plot TCN Vs P | P-FLAG

Isotherm Temperatures  
P(kPa) TCN(W/mK) Flag

196	101.325 0.1597 0
203	101.325 0.1568 0
213	9000 0.152 7
220	30000 0.1598 7
223	40000 0.161 7
223	101.325 0.1505 1
233	50000 0.1609 7
243	35000 0.1533 3
248	20000 0.156 7
253	500 0.153 7
263	10000 0.155 7
273	25000 0.1579 7
283	30000 0.1608 7
293	40000 0.1639 7
298	101.325 0.1533 0
303	12000 0.1522 7
307	600 0.1483 3
313	30000 0.1582 7
321	40000 0.1615 7
323	101.325 0.1498 0
325	20000 0.1553 7
333	9000 0.145 7
343	50000 0.1582 7
345	101.325 0.1433 1

Linear Reg of TCN-P(>30)  
Parabolic Reg  
a  
b  
c

Linear Reg of TCN-P(>30)  
Linear Reg  
m  
c

Regression > 101  
Avg gradient of all TCN vs P graphs for current DDB num.

Lin Reg of TCN-T for P = 1atm  
Lin Reg  
m: -3.52E-04  
c: 0.2243657

Thermal Conductivity (W/mK)

Temperature (K)

- The DDB (Artist)
- Microsoft Office
- SQL & ADO

# Future Work

- ✿ Select a regression model and determine empirical parameters for all compounds
- ✿ Evaluation & definition of structural groups
- ✿ Regression for structural or bond contributions
- ✿ Testing using a test set of thermal conductivity data
- ✿ Optimisation of structural or bond contributions

# Summary

- ◆ Importance of prediction methods
- ◆ Thermal Conductivity
- ◆ Review of available prediction methods
- ◆ Evaluation of methods
- ◆ Filtration and validation of data from DDB
- ◆ Software
- ◆ Work left to be done

# Acknowledgements

- ✿ Prof. D. Ramjugernath from UKZN
- ✿ Prof. J. Rarey and Prof. J. Gmehling from Oldenburg University
- ✿ Bruce Moller and Eugene Olivier
- ✿ DDBST Software and Separation Technology
- ✿ South African NRF and German DLR (BMBF)