

Edit Mixture Data

Entering Experimental Data of Mixtures in the Dortmund Data Bank

DDBSP – Dortmund Data Bank Software Package



DDBST Software & Separation Technology GmbH

Marie-Curie-Straße 10

D-26129 Oldenburg

Tel.: +49 441 361819 0

Fax: +49 441 361819 10

E-Mail: support@ddbst.com

[Web: www.ddbst.com](http://www.ddbst.com)

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1. Introduction

1. Supported Data Banks

EditMixtureData is the single editor for the mixture data banks in the Dortmund Data bank (DDB).

The supported data banks are currently:

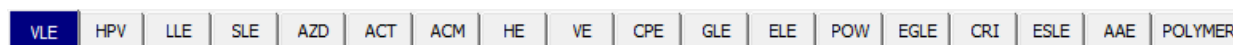


Figure 1 Data Bank Pages

Data Bank	Description
VLE	Vapor-liquid equilibria for high boiling substances (above 0°C)
HPV	Vapor-liquid equilibria for low boiling substances (below 0 °C)
LLE	Liquid-liquid equilibria (miscibility gaps)
SLE	Solid-liquid equilibria
AZD	Azeotropic and zeotropic data
ACT	Activity coefficients at infinite dilution for pure solutes in pure solvents
ACM	Activity coefficients at infinite dilution for pure solute in binary solvents
HE	Heats of mixing
VE	Excess volumes (volume effects of mixing), densities and volumes of mixtures
CPE	Excess heat capacities and heat capacities of mixtures
GLE	Gas-liquid equilibria (gas solubilities)
ELE	Vapor-liquid equilibria for electrolyte containing systems
POW	Octanol/Water partition coefficients
EGLE	Gas-liquid equilibria (gas solubilities) in electrolyte containing systems
CRI	Critical data of mixtures
ESLE	Solid-liquid equilibria for electrolyte containing systems (salt solubilities)
AAE	Adsorbent/adsorptive equilibria
POLYMER	Polymer data

2. Working Mode

EditMixtureData is directly working on the data banks – no intermediate files are involved. This is contrary to EditPureData which primarily works on files.

Additionally it is possible to load and store intermediate files in binary or formatted formats. These formats are not intended for editing in external editors.

This working mode has some advantages and some disadvantages:

- No extra step is needed for updating the data bank
- Original data in original units are not available any more – we recommend to use a spreadsheet program (Microsoft Excel or OpenOffice Calc for example) for keeping original data. EditMixtureData can easily exchange data with these programs by copy and paste.

3. DDB Structure

As this document uses some terms referring to the structure of the DDB some general information about the structure is given here.

There's a global *configuration* that contains information used by all DDBSP applications. This configuration is stored either on drive C : (a file named DDB.CFG) or in the Windows registry.

Sample configuration:

```
@DDBPTH=F:\DDB\
@DDBPRV=F:\PRIVATE\
@DDBSCR=F:\SCRATCH\
@DDBTRC=ON
```

A sample contents of this file is shown here:

The entry DDBPTH points to *public folder*, the directory holding *public data* (provided by DDBST GmbH).

The entry DDBPRV is set to the location of the *private folder* (containing data maintained by customers only).

The entry DDBSCR determines the *clipboard directory*, which will be used by DDBSP applications to exchange input and output files as well as temporary files. Sometimes files created there contain additional information not displayed by the programs.

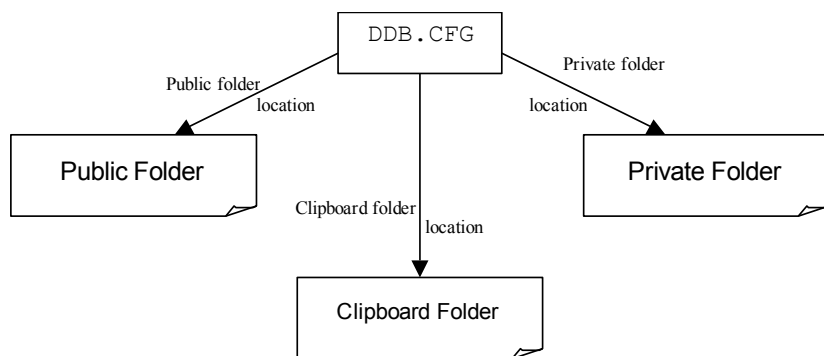


Figure 2 Rough Structure of the DDB folders

Figure 3 shows in more detail how the mixture data banks are organized within the DDB. Published experimental data refer to *components* (defining the system) and *literature data* (representing the paper containing the data). The DDB contains a large component list (more than 19000) as well as more than 70000 literature datasets. These are physically organized outside the actual mixture data banks.

Experimental data provided by DDBST will refer to public components and public literature only. Customer's data may also refer to private (user-defined) components and private literature.

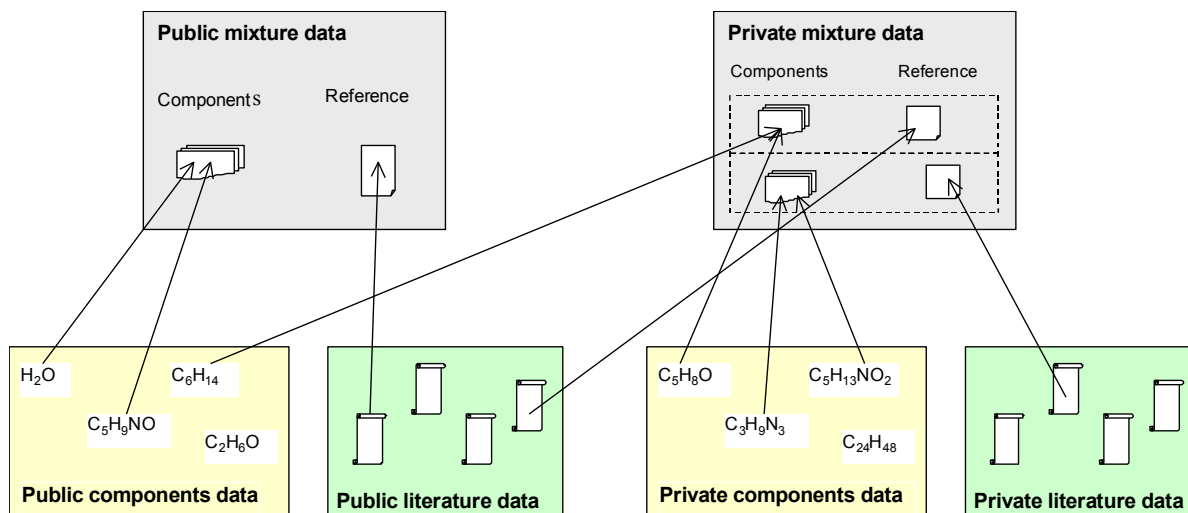


Figure 3 Organization of Literature Data

As everything denoted as *public* is provided by DDBST and all (unreported) changes done to public data will be

eradicated with an update this document will focus on the maintenance of *private data*.

4. Organization of Literature Data

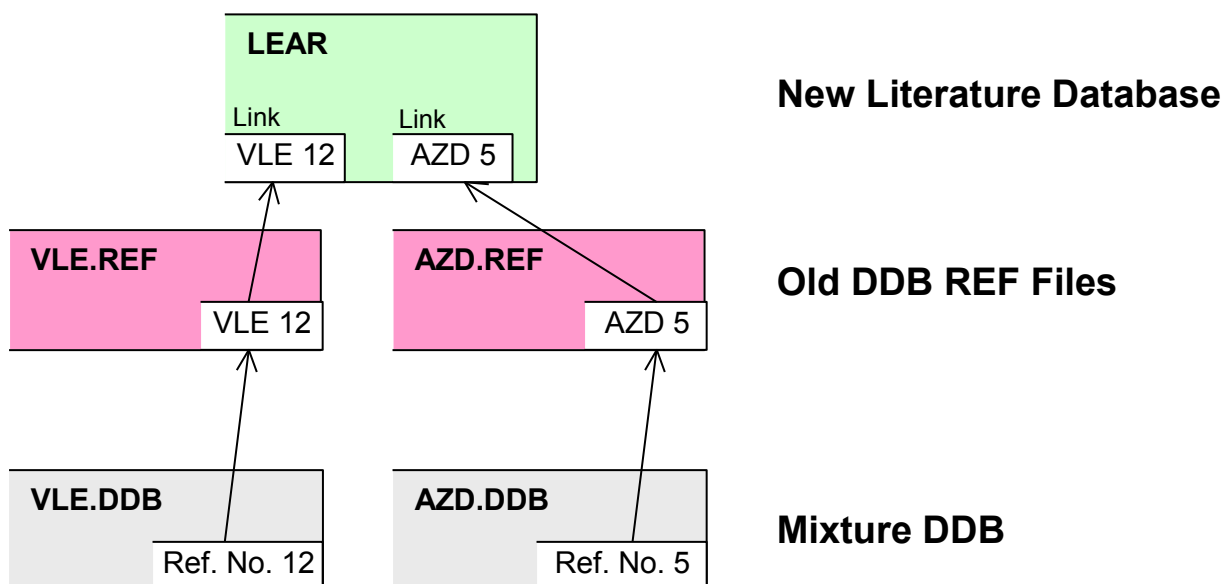


Figure 4 Organization of Literature Data (Example)

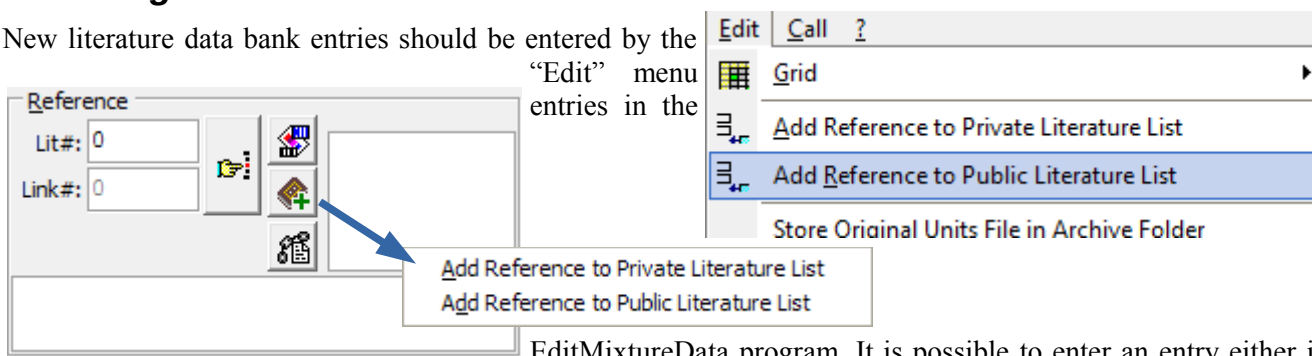
As an example we assume an article that contains both VLE and AZD experimental data. The evaluation will result in two data bank entries: one in VLE.DDB and one in AZD.DDB. The reference number organization in the mixture data bank uses different lists of references for every data bank. So this will result in two entries in the old references data bank (VLE.REF and AZD.REF, both reference lists start from reference no 1). But these files exist for backward compatibility only. Instead the article will be referenced only once inside LEAR (representing both REF entries). As there is a reference number stored in a mixture data set there must be an association between a LEAR data set and all reference numbers belonging to it. This is done by adding a so-called *link* to the LEAR data set. In this example a link to REF VLE 12 and REF AZD 5 will be added.

2. Data Input

Since the data sets only contain references to components and literature information it is necessary to add component and literature information prior to the experimental data.

1. Adding New Literature

New literature data bank entries should be entered by the “Edit” menu entries in the




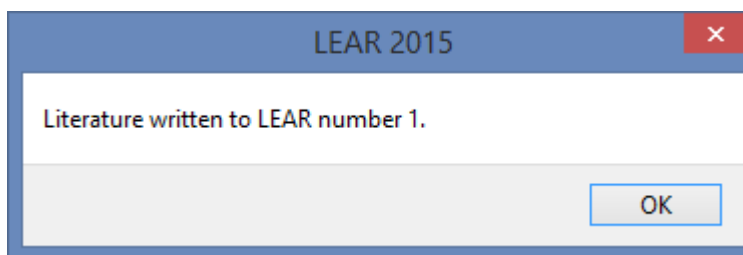
EditMixtureData program. It is possible to enter an entry either in

the private or the public literature data bank.

Figure 5 New Entry in the Literature Database

This literature sheet can be left completely empty but we recommend to enter at least a title and an author. This makes it easier to identify the entry in the editor.

After saving this sheet () a saving confirmation dialog is displayed. As shown above the database needs additionally a special “link” to the current database.



This link is created after confirming the question. This link is needed for internal referencing.

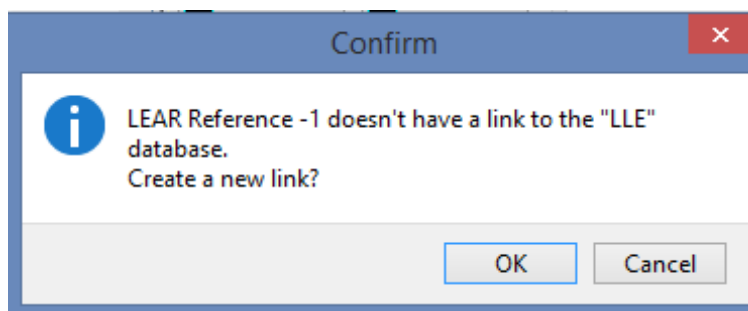


Figure 6 Link Creation

Now we have a valid and complete new reference which now shown in the “Reference” block in EditMixtureData.



Figure 7 Reference Information

2. Adding a Private Component

The components are edited in the program “Component Editor” which can be called from the “Call” menu.

We will quickly show the creation of a new private component, whereas a complete description of the program can be found elsewhere.

The program always starts displaying the public component no. 1 – Acetaldehyde. Appending a new data set is done by the



button.

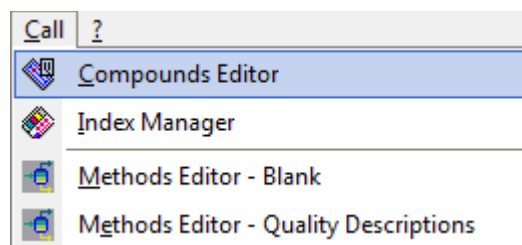


Figure 8 Add Component

In the dialog (Figure 8) it is necessary to fill the green fields (2 names, formula, molecular weight) before the component can be added either to the public or the private component list.

If a private component list does not yet exist it will be created automatically.

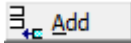

The additional fields are only necessary for calculations – EditMixtureData only needs the component number to proceed. The molecular weight can become important if conversions from mole fractions to weight fraction are needed.

3. Common Options of EditMixtureData

1. Searching Components

Figure 9 Component Box

The edit field below the 'Salt' button allows to type DDB component numbers if they are known.

Pressing the  **Add** or the  **Salt** button invokes the ComponentSelection program either for 'normal' components or for salts. In both cases it is possible to switch to the other list and to the other types of components which are adsorbents and polymers for the data banks containing salts (ELE, else), adsorbents (AAE) and polymers.



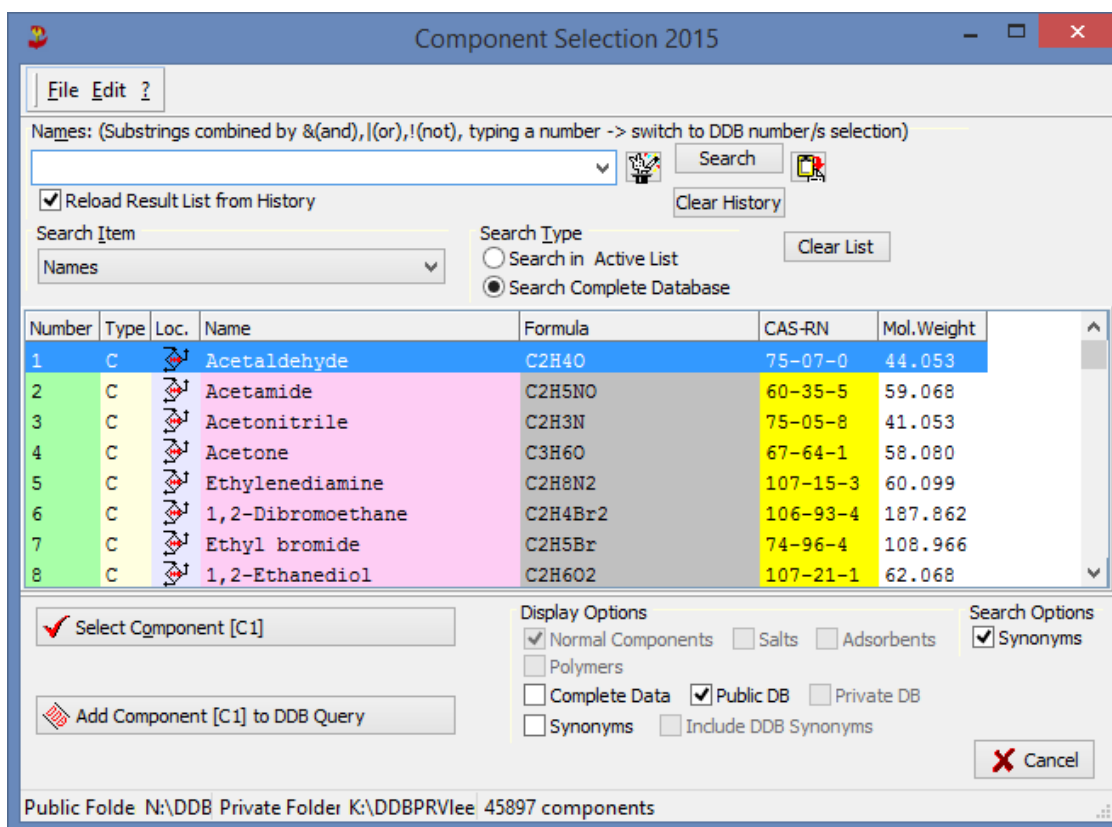


Figure 10 Component Selection

After selecting two normal components and a single salt the component box looks like this:

4	C	DDB	Remove	Acetone
44	C	DDB	Remove	1-Chloro-2-methylpropan-2-ol
7	S	DDB	Remove	Barium bromide

The green column contains the component numbers, the light yellow column contains a “C”, a “S”, or an “A” for the three types normal components, salts/electrolytes, and adsorbents.

The **Remove** button removes the component/salt/adsorbent and the pink column contains the name.

Depending on the currently selected data bank type not all components will be used for input. All unused components will be marked by a gray color.

4	C	DDB	Remove	Acetone
44	C	DDB	Remove	1-Chloro-2-methylpropan-2-ol
7	S	DDB	Remove	Barium bromide

The order of the components can be changed by drag-and-drop (moving the mouse while keeping the left mouse key pressed).

4	C	DDB	Remove	Acetone
44	C	DDB	Remove	1-Chloro-2-methylprop
7	S	DDB	Remove	BARIIUM BROMIDE

Figure 11 Start

4	C	DDB	Remove	Acetone
44	C	DDB	Remove	1-Chloro-2-methylprop
7	S	DDB	Remove	BARIIUM BROMIDE


Figure 12 Move/Drag

4	C	DDB	Remove	Acetone
7	S	DDB	Remove	BARIIUM BROMIDE
44	C	DDB	Remove	1-Chloro-2-methylprop

Figure 13 Drop/Result

The same reordering can be performed by the arrow keys <▲> and <▼> while holding down the Control (Ctrl,Strg) key.

2. Searching Literature

If a literature is already available it can be searched via the search buttons () in the “Reference” block (see Figure 7). The search query dialog contains two search modes, a quick search mode and an advanced search mode.

We will show a quick search for the author “Lemmon” and the year 1999 (see Figure 14).

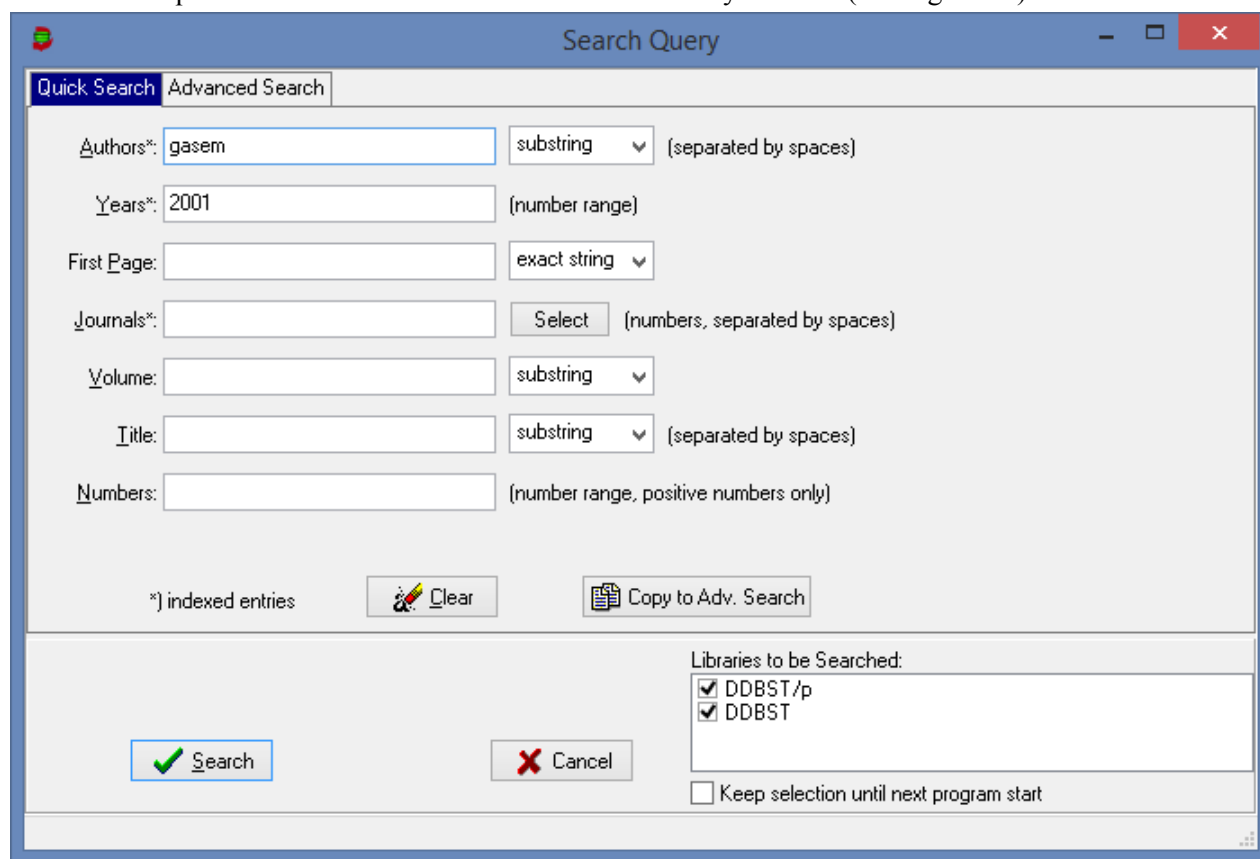

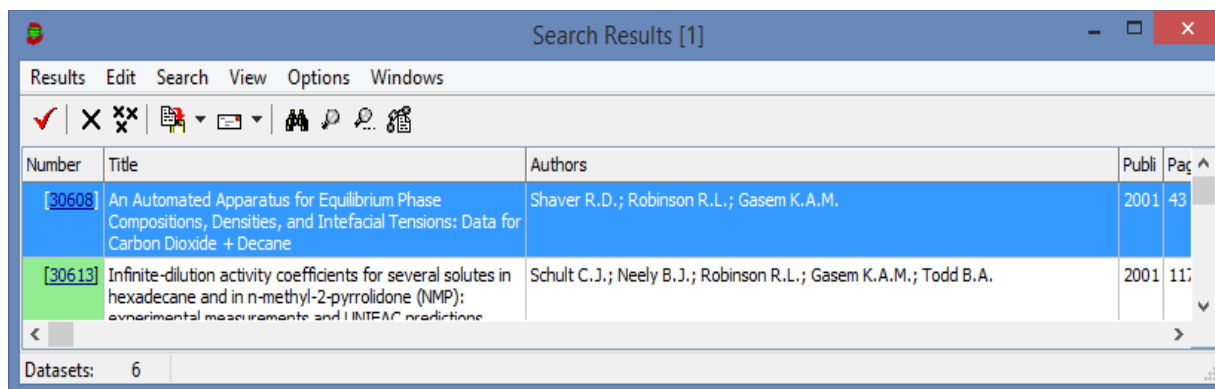


Figure 14 Quick Search Query

After selecting the single found data set () EditMixtureData checks if the selected data set contains an appropriate link for the current data bank, and creates a link if necessary.

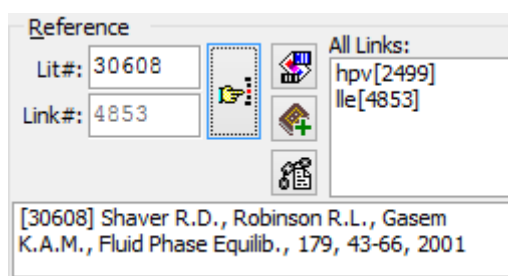


Number	Title	Authors	Publi	Pag
[30608]	An Automated Apparatus for Equilibrium Phase Compositions, Densities, and Interfacial Tensions: Data for Carbon Dioxide + Decane	Shaver R.D.; Robinson R.L.; Gasem K.A.M.	2001	43
[30613]	Infinite-dilution activity coefficients for several solutes in hexadecane and in n-methyl-2-pyrrolidone (NMP): experimental measurements and UNIFAC predictions	Schult C.J.; Neely B.J.; Robinson R.L.; Gasem K.A.M.; Todd B.A.	2001	11

Datasets: 6

Figure 15: Search Result

The selected reference is displayed in the “Reference” block. The LEAR (main literatur) number, the REF (link for the currently selected database) number, all available (other) links, and a short string showing the author, the journal, the pages and the year.



Reference

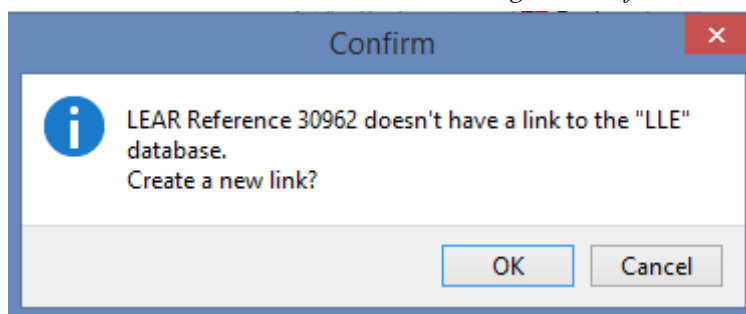
Lit#: 30608

Link#: 4853

All Links:
hvp[2499]
lle[4853]

[30608] Shaver R.D., Robinson R.L., Gasem K.A.M., Fluid Phase Equilib., 179, 43-66, 2001

Figure 16 Reference Block



Confirm

i LEAR Reference 30962 doesn't have a link to the "LLE" database.
Create a new link?

OK Cancel

If a LEAR or a link number is known it can be directly typed in the LEAR or REF edit fields. After accepting the input by pressing enter EditMixtureData shows the details.

3. Plot Options

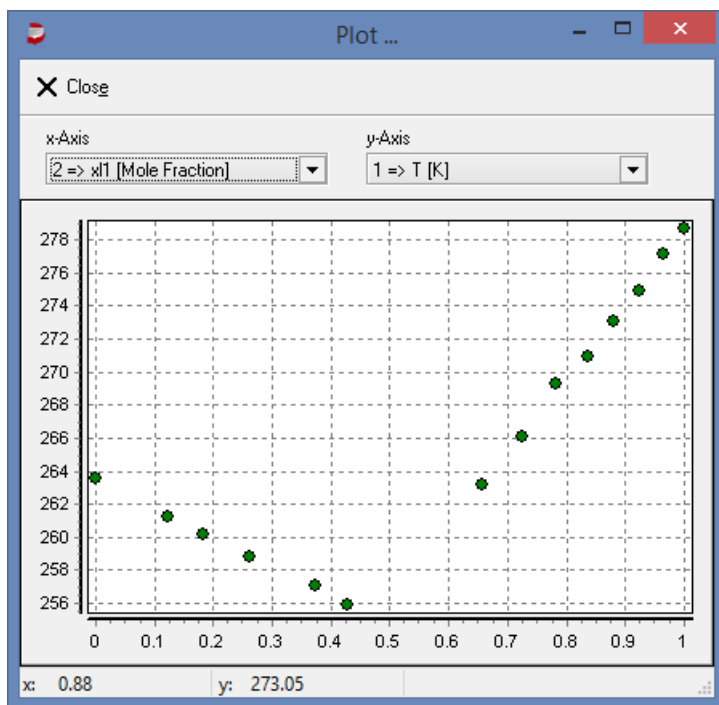



Figure 17 Quick Plot

To receive a quick overview about the just edited data the quick plot option

( Quick Plot) can be selected.

The integrated plot of EditMixtureData only allows to plot two data columns against each other.

The selected columns can be changed by the comboboxes:

x-Axis
2 => x1 [Mole Fraction]

y-Axis
1 => T [K]

The status bar displays the x,y data for the point which is next to the mouse cursor indicated by a rubber band.

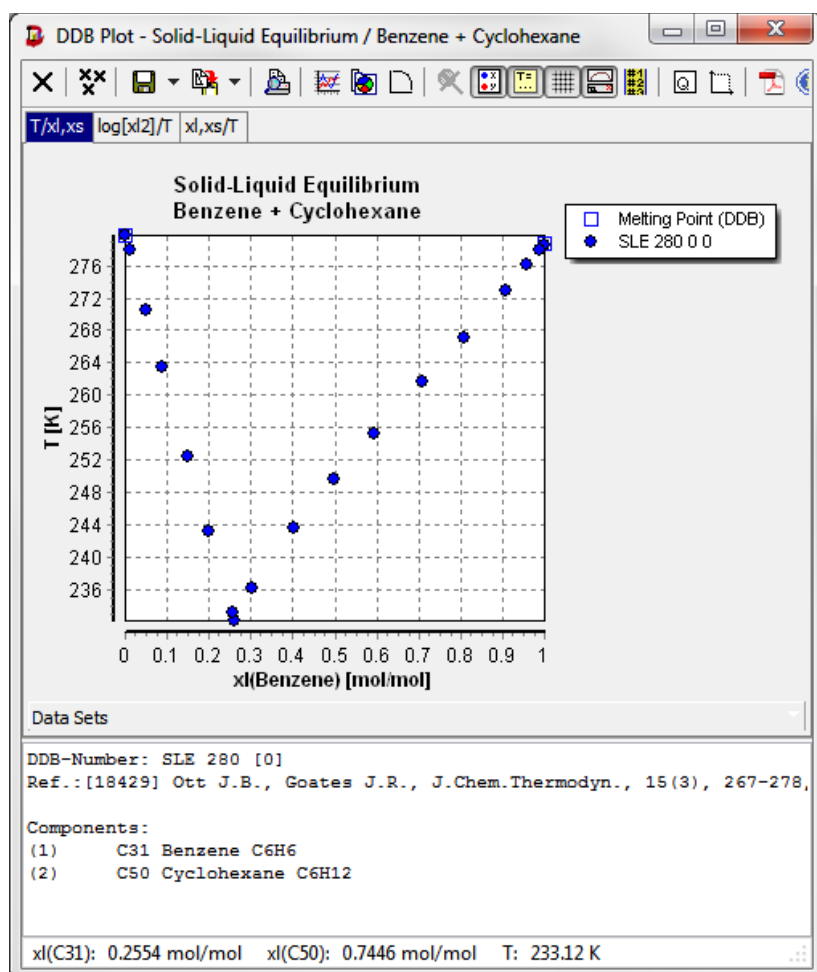
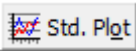
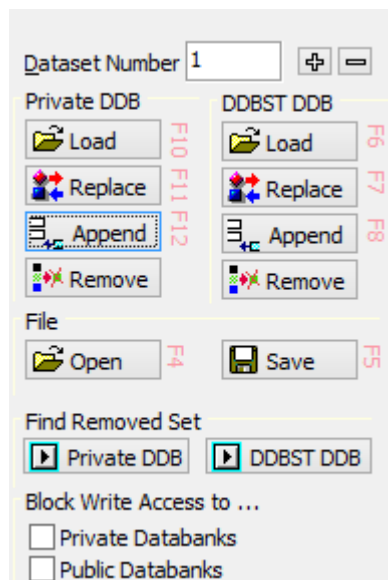


Figure 18 Standard Plot

The standard plot  calls the plot program normally used. This plot program automatically selects the generally used plot types for the current datasets.

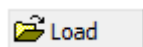
It displays the components, the reference and also the point's data which is next to the mouse cursor in the status line.

4. Loading from and Storing to the DDB



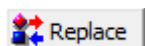
EditMixtureData stores and reads datasets directly from the data banks.

There are two groups of buttons, one for the private (customer's) and the public (DDBST) database.

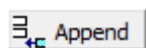


To change an existing dataset (or to use its data as basis for a new dataset) use the appropriate *Load* button. Before loading a dataset from the (private) DDB be sure that the correct data bank tab is selected. “Load” will always fetch the dataset specified in the Dataset Number 1 field.

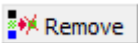
There are two possibilities to put a datasets directly into the (private) DDB: A dataset can either replace an existing dataset or it can be appended to the data bank, leaving all other datasets untouched.



The *Replace* option is intended for corrections of existing datasets. But it is, of course, possible to replace an existing dataset by a complete new dataset. Before replacing a dataset a confirmation dialog will prevent the user from accidentally messing up the data bank and additionally every overwritten data set is saved in a “Roll Back” file.



To add a new dataset to the (private) DDB use the *Append* option. If the data bank does not exist so far, a brand new will be created and the current dataset will be added as first (and only) dataset. Anyway, the *Dataset number* field will be updated to the new number.

The last button () deletes the dataset specified in the Dataset Number field from the current database.

Previously removed sets can be retrieved by the button in the “Find Removed Set” group.

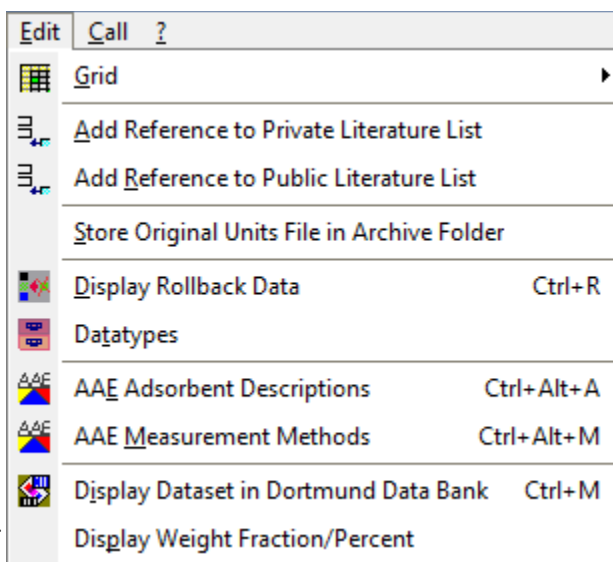
5. Restoring Overwritten Data Sets

Every replaced and overwritten data set is saved in a roll-back file. These saved data sets can be restored from this list.

The dialog displays the entire list of overwritten sets which can become extremely long. For example the list here at the DDBST contains over 19000 entries collected over many years for all data banks.

The view on the list can be restricted to single data banks, single users, single locations (private or public DDB) and, the most effective filter, the data set number.

Data bank, data set number, location, data and time of overwriting, and the user is displayed.



The **Restore** button allows to restore the selected data set. The data set which is now replaced by the old data set will be stored in the roll-back file.

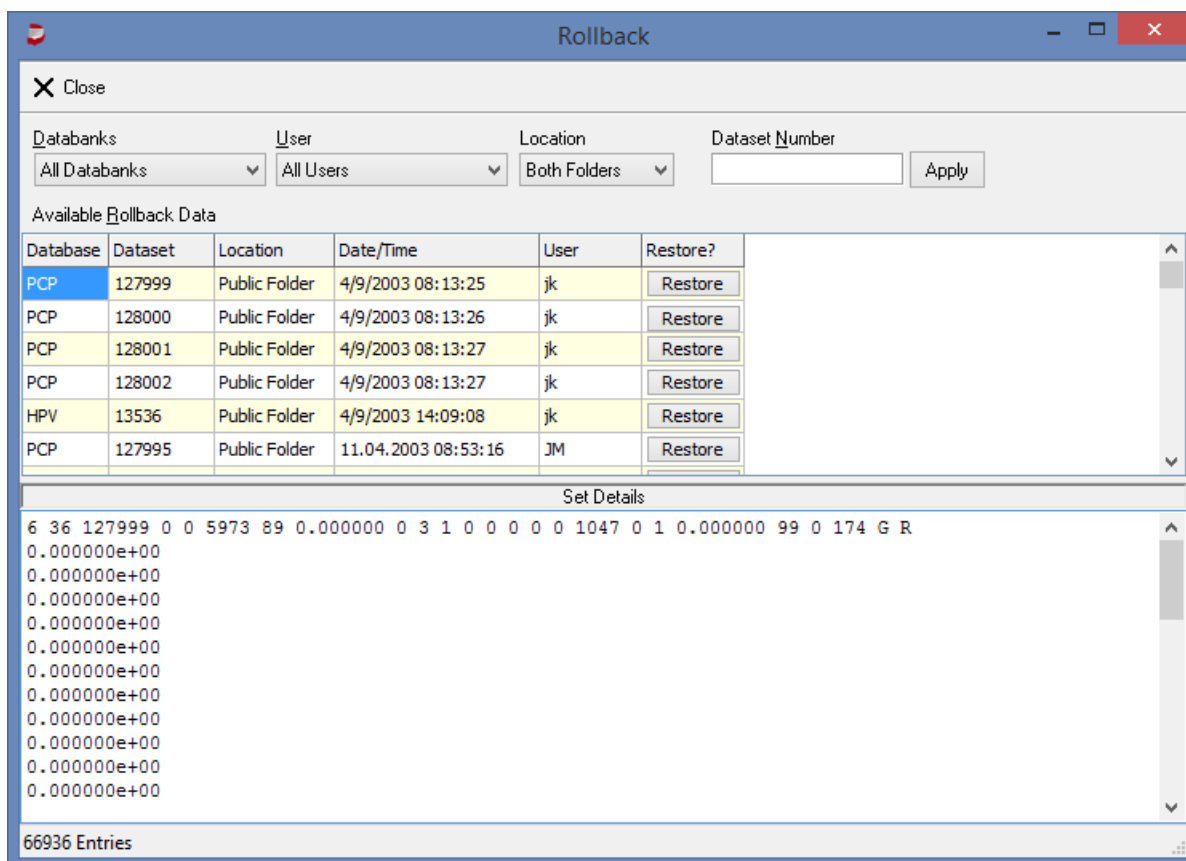
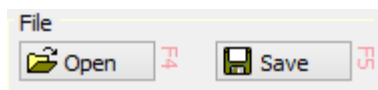


Figure 19: Rollback Dialog

6. Import/Export Options



Datasets can be written or loaded from interface files in several formats. All file formats are currently proprietary formats of DDBST.

- **Query Result File**
This is almost the old “Free Formatted DDB Interface File” but extended by a database information. This format should be used wherever possible. The extension is “.qr”.
- **Binary DDB Interface File**
This is a legacy file format normally used by old DOS programs in the DDB software package.
- **FORTRAN 77 Formatted DDB Interface File**
This is another legacy file format. This file format can only be read. It contains text information but is formatted very dense which makes it hard to read.
- **Data Handling Program File**
This format is very similar to the FORTRAN 77 formatted file and has been used by some DOS programs.

This format can only be read.

- Free Formatted DDB Interface File
This is the new standard file format.
- Original Units File
For some databases it is possible to store the typed data in original units.

Several of these files can contain multiple datasets. If such a file is loaded a special dialog displays the complete list of found datasets.

This dialog shows the database, the given number of data points, the component list (as DDB numbers) and the reference number (links).

It is possible to edit the single dataset (**Edit**) or to append all sets either in the public or private data bank.

If the data bank contains removed data sets they are listed in the “Empty Sets” list and will be overwritten if the “Fill Empty Sets” box is checked.

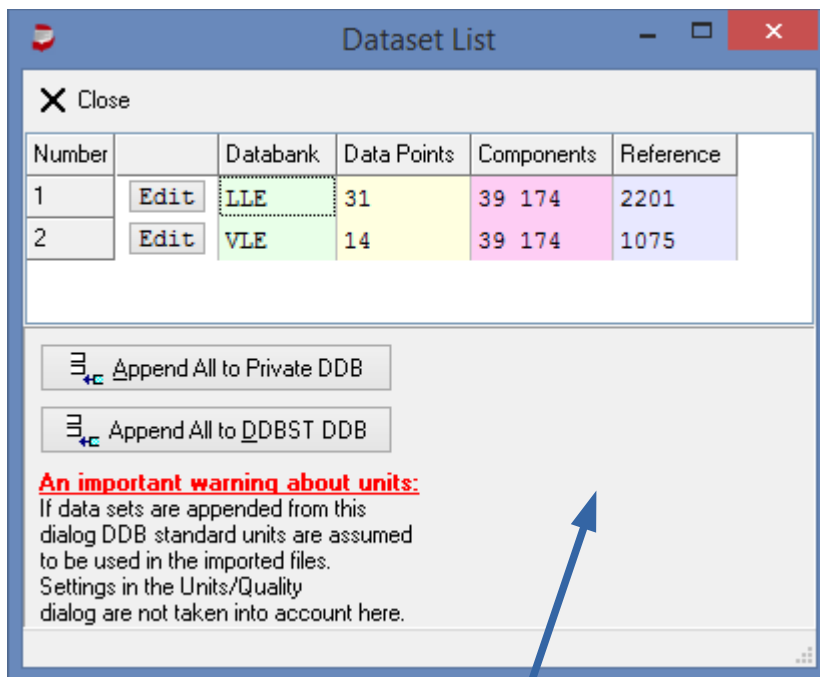
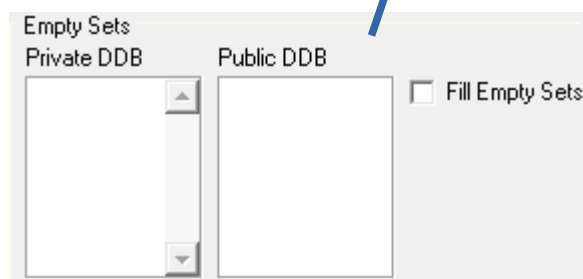


Figure 20 List of loaded datasets



Remark: VLE/HPV data saved in the same binary format having the same file extension. So the program can not differ between these two types. The user has to select the right data bank tab before loading data of one of these two types. All other binary types are recognized automatically.

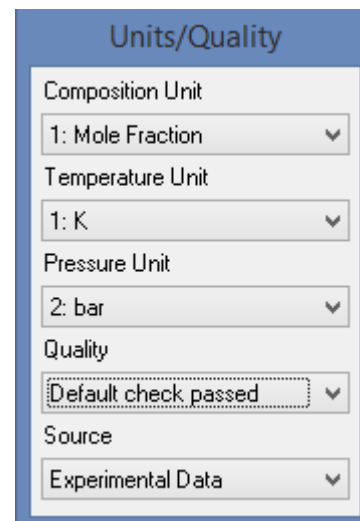
Select the **Save** button to store a dataset as file on disk. Depending on the extension selected in the file open dialog the file is either stored in text or in binary mode. If the data is stored as binary file the data type is automatically recognized by Mixture Properties using the file extension. This is also true for the “.qr” query result files.

7. Units and Qualities

The input units dialog – also containing selection boxes for data quality and source descriptions – is a floating dialog which is always displayed on top of the main window.

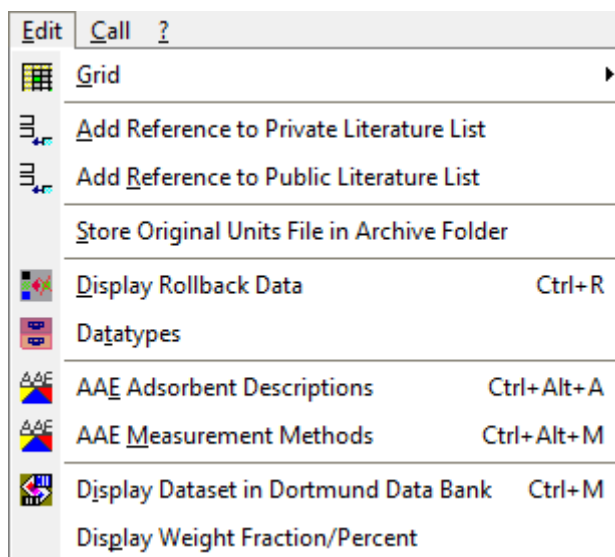
When switching to another database the program displays the standard units initially.

Some databases have their own unit description (GLE for example). In this case the selection box is grayed out.



8. Display Data Set in Dortmund Data Bank

This function transfers the currently edited data set to the main Dortmund Data Bank program without storing it to the data bank before-hand.



This allows to use the entire functionality of the Dortmund Data Bank program like extended plotting, estimation, fitting, exporting, and more.

4. Input Options Separated by Data Bank

It is recommended to select the appropriate data bank before selecting components or literature. The selected literature data set must be prepared for use with the data bank.

1. VLE – Vapor-Liquid Equilibrium Data of Normal Boiling Substances

Select the VLE tab above the edit field to enter vapor-liquid equilibrium data for systems containing components with a boiling point above 0°C. The number and heading of the input columns will be set automatically by (de-) selecting of components and the other options described below.

The maximum number of components in a system is four.

Data Type				
1	x	y	P	T=constant
2	x	y	T	P=constant
3	x		P	T=constant
4	x		T	P=constant
5	x	y		T=constant
6	x	y		P=constant
7		y	P	T=constant
8		y	T	P=constant
9	x	y	P	T

As first configuration in the data sheet the desired data type has to be selected.

There are currently nine data types defined. They specify which phases are given and if there's a constant pressure or temperature for the whole data set.

The liquid phase is denoted 'x' and the vapor phase is denoted 'y'.

For "Iso" values 1 to 8 a constant condition is given for all data points. So a value and the value's unit has to be given.

Constant

 °C

Constant

 mmHg

Data Type				
9	x	y	P	T

☒ x ☒ y ☒ T ☒ P
 (Almost) Constant:
☐ x ☐ y ☐ T ☐ P ☒ none

A composition is always given in mole fractions. The value for the last component in the system can (and has to) be omitted.

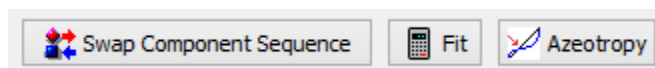
If no constant temperature or pressure given for the whole dataset, then "Iso" 9 has to be selected. With this selection a second "Non-Constant Unit" field will appear. It is also possible to specify which measured quantity is really given. In this case a special option can be used to define a (pseudo) constant column. By this it is allowed either to specify a constant value for one phase or to state that the deviation of the values of one column is small.

Although any data classified as type 1-8 can be seen as special type 9 data it is recommended to avoid the usage of this data type. The data types one or two should be used as often as possible, because most calculation routines are optimized for them.

The VLE editor allows also performing consistency tests, swapping components, fitting g^E model parameters (for test purposes here), and starting the evaluation if the entered VLE data show azeotropic behavior.

Consistency Tests

Area Point to Point



2. HPV – Vapor-Liquid Equilibrium Data of Low Boiling Substances

Select the HPV tab to edit vapor-liquid equilibrium data for systems containing components with boiling points below 0 °C. The options are mostly the same as for VLE data. Only the pressure and temperature units used in the DDB are different.

3. LLE – Liquid-Liquid Equilibrium Data

To edit liquid-liquid equilibrium data (miscibility gap information) the LLE tab has to be selected. It is recommended to select the desired system and literature data set before configuring the data sheet itself. The selected literature data set must be prepared for use with the LLE data bank. The maximum number of components in a system is five.

The input of this kind of data differs much from editing e.g. VLE data. For example no predefined data types are used. Instead the table entries are selected directly.

All the settings described below (and the number of components) control the number of columns in the data grid.

Table Entries

☐ Pressure

☒ Temperature

☐ Vapor Composition

☒ Liquid Composition

The *Table Entries* specify the measured quantities. It is not possible to specify a constant environment condition directly. In this case the constant value has to be entered in any row of the data grid.

Pressure Type

☐ Equilibrium (VLE)

☐ non-Equilibrium

☒ Not specified

If a pressure is given, then the *Pressure Type* has to be specified. If unsure *Not Specified* should be selected.

Number of Phases

2

If there are three or more components in the system, then up to three liquid phases may be specified in the field *Number of Phases*. A possibly given vapor phase is not counted here.

4. SLE – Solid-Liquid Equilibrium Data

Select the SLE tab above the edit field to enter solid-liquid equilibrium data (solubility information) and add it to the SLE data bank. The number and heading of the input columns will be set automatically by (de-) selecting of components and the other options described below. It is recommended to select the desired system and literature dataset before configuring the data sheet itself. The selected literature dataset must be prepared for use with the SLE data bank. The maximum number of components in the system is four.

The desired data type can be selected in the *Dataset Entries* field. There are currently nine main types defined which are described below.

Dataset Entries

1 - T xl (P)

1 - T xl (P)

2 - T xs (P)

3 - T xl xs (P)

4 - T xl xs y (P)

5 - P xl (T)

6 - P xs (T)

7 - P xl xs (T)

8 - P xl xs y (T)

9 - T P xl xs y

‘T’ denotes a temperature is given. ‘xl’ is an abbreviation for mole fraction in the liquid phase. ‘xs’ means mole fraction in the solid phase. ‘y’ is for mole fraction in gas phase. A symbol is listed in brackets denotes an environment condition valid for all data points.

Constant

101

For the data types 1 to 8 a constant condition is given for all data points. So a constant value and the value’s unit both have to be specified.

Dataset Entries

9 - T P xl xs y

Available columns:

☒ P ☒ T ☒ xl ☒ xs ☒ y

(Almost) Constant:

☐ P ☐ T ☐ xl ☐ xs ☐ y ☒ None

There is a reserved data type (9) to specify that neither a temperature nor a pressure as constant environment condition is given. This data type 9 can also be used if P/T is unknown for any data point. Selecting this data type will make up to two *Non-Constant Unit* fields to appear. Again, it is possible to define one (almost) constant column.

Although any data classified as type 1-8 can be seen as special type 9 data it is recommended to avoid the usage of this data type. Most of the plot and calculation routines will only work with the data types 1-8.

SLE Type

unknown type

unknown type

eutectic system

part.misc. in liquid phase

comp. form stable compound

peritectic system

completely misc. in solid phase

dystectic system

part.misc. in solid phase

part.misc. peritectic solid phase

comp.inmisc. in liquid phase

solubility of polymorph species

Several classifications are available in the *SLE Type* drop-down list. If unsure “*unknown type*” should be selected.

5. AZD – Zeotropic/Azeotropic Data

To edit azeotropic/zeotropic data the AZD tab has to be selected. It is recommended to select the desired system and literature data set before configuring the data sheet itself. The selected literature dataset must be prepared for use with the AZD data bank. The maximum number of components in the system is four.

Any data set in the AZD data bank contains just one data point. It is possible to specify a temperature, a pressure and (if an azeotrope exists) a composition. If no temperature, no pressure or no composition is given, then the corresponding option should be unchecked.

Furthermore some additional information can be specified. These will be described below.

Type of Azeotrope: If beyond the fact that there’s an azeotrope no further information is known, then this field should be left empty.

Type of Azeotrope

<input type="checkbox"/> 0 - Unspec. Azeotr.	<input type="checkbox"/> 32 - Saddle
<input type="checkbox"/> 1 - none	<input type="checkbox"/> 64 - 2azd
<input type="checkbox"/> 2 - hom	<input type="checkbox"/> 128 - MisGap
<input type="checkbox"/> 4 - het	<input type="checkbox"/> 256 - supercrit
<input type="checkbox"/> 8 - Pmax	<input type="checkbox"/> 512 - 3 liq.phases
<input type="checkbox"/> 16 - Pmin	<input type="checkbox"/> 1024 - Sep.Fac.~1.0

To specify that there's *no azeotrope* (maybe under certain environment conditions) for the given system *none* can be selected. Of course, this will exclude a detailed specification of the azeotrope described below.

Selecting *Pmax* resp. *Pmin* will specify a pressure maximum resp. minimum azeotrope, where *Saddle* specifies a saddle point.

If there exists a miscibility gap under the given environment conditions *MisGap* should be selected.

Enabling *2azd* specifies the existence of two azeotropes.

Select *supercrit* if one of the components is supercritical under the given conditions.

3 liq.phases specifies the co-existence of three liquid phases.

Sep.Fac.~1.0 is a qualitative statement that the separation factor is near 1.

Measurement Method

<input checked="" type="radio"/> n.a.
<input type="radio"/> Distillation
<input type="radio"/> Phase Equilibrium

Measurement Method: If the measurement method is unknown or none of the listed, then the first entry in the list should be selected.

Otherwise select *Distillation* if the information has been obtained by distillation or choose *Phase Equilibrium* (based on phase equilibrium measurements).

☐ Reaction

Reaction: If the system is reactive select *Yes*, otherwise *No* should be chosen.

Evaluation by

<input type="checkbox"/> TC OL. (Checked) or author
<input type="checkbox"/> table
<input type="checkbox"/> graph
<input type="checkbox"/> VLE
<input type="checkbox"/> fitted
<input type="checkbox"/> calculated

Evaluated by: The *Evaluated by* field contains information about the person and how the data has been gained. Leaving the first entry unchecked means the evaluation has been done by the author.

If the data has been published as table or in a graphical representation choose the corresponding checkbox.

Select *VLE* if the information has been provided as VLE data. If the information has been extracted by fitting to published VLE data, then *fitted* should be selected. If furthermore the information has been calculated using the fitted parameters obtained by the published VLE data, then *calculated* should be activated.

Quality

not given ▼
not given
++++
+++
++
+
+/-
-
--

Quality: The data quality can be stated. If the quality information is not available at input, *not given* should be selected. Otherwise choose one between the ranking *very bad* (----), medium (+/-) and *very high* (++++).

The precision of a measured value can be specified using the according drop-down list. '≡' denotes exact, '<' denotes lower than, '≈' denotes about, '>' denotes higher than and '?' denotes a dubious value.

6. ACT – Activity Coefficients at Infinite Dilution (Binary Systems)

Select the ACT tab to edit activity coefficients at infinite dilution data for binary systems.

The first listed component of the system will be interpreted as *solute*, the second component will be assumed to be the *solvent*. In addition the selected classification will be displayed in the bottom area of the edit field.

Besides the activity coefficient at infinite dilution a temperature has to be specified.

- 0 - prediction
- 1 - GLC (with gas phase correction)
- 2 - GLC (without gas phase correction)
- 3 - LLC
- 4 - ebulliometry
- 5 - dilutor
- 6 - static method
- 7 - GLC (not specified)
- 8 - GLC (relative method)
- 9 - other techniques
- 10 - calculated from Henry coefficients
- 11 - GLC (mean values at average temperature)
- 12 - GLC (non steady state method)
- 13 - Rayleigh distillation method
- 14 - calculated from solubility data
- 15 - calculated from phase equilibria

It is also possible to state the *Measurement Method* that has been used to obtain the activity coefficient. If the method is unknown or none of the listed, then *other techniques* should be selected.

There are several predefined experimental methods like GLC (including specializations), LLC, ebulliometry, dilutor and the Rayleigh distillation method.

If the value has been calculated either the general prediction or the calculated from Henry coefficients can be selected.

If the data has been published in the *DECHEMA Data Series* the corresponding information can be stored using the *Part* and *Page* edit field. Zero in either field means “not classified”.

Currently every data set contains only one data point. It is though possible to type more lines. Every line typed here will become a separate data set. It is not possible to load the typed line simultaneously.

7. ACM – Activity Coefficients at Infinite Dilution (Ternary Systems)

Select the ACM tab to edit activity coefficients at infinite dilution data for ternary systems.

Here the *third* listed component of the system will be interpreted as *solute*. The first two components will be assumed to be the *solvent system*. The right selection can be validated inspecting the *Solvents* and *Solute* field where the DDB component numbers will be displayed.

For any data point a temperature, the composition (for the first component) and the activity coefficient itself has to be specified.

- 0 - Prediction
- 1 - GLC (with gas phase correction)
- 2 - GLC (without gas phase correction)
- 3 - LLC
- 4 - ebulliometry
- 5 - dilutor
- 6 - static method
- 7 - GLC (not specified)
- 8 - GLC (relative method)
- 9 - other techniques
- 10 - calculated from Henry coefficients
- 11 - GLC (mean values at average temperature)
- 12 - GLC (non steady state method)
- 13 - Rayleigh distillation method

For the whole dataset a *measurement method* can be specified.

0 not specified

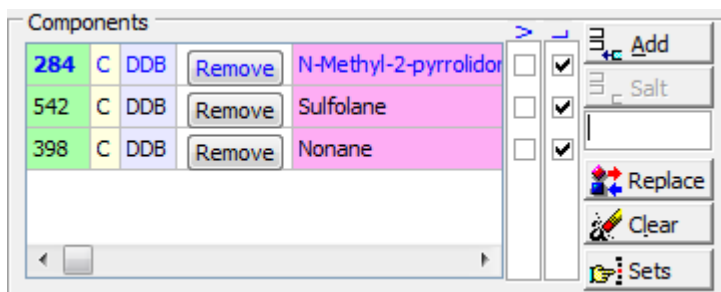
- 1 ++++
- 2 +++
- 3 ++
- 4 +
- 5 +/-
- 6 -
- 7 --
- 8 ---
- 9 ----

In addition the quality of the data (whole dataset) may be specified. If unsure select *not specified*.

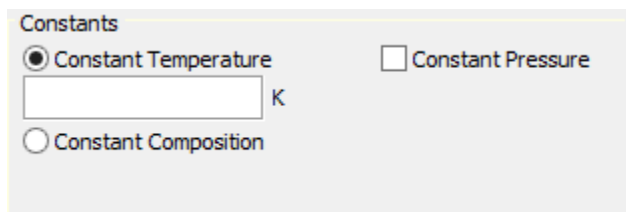
As for AZD otherwise a ranking from *very bad* (----) over medium (+/-) up to *very high* (+++++) is possible.

8. HE – Heats of Mixing Data

Excess enthalpy data (heats of mixing) can be edited using the HE tab. The maximum number of components in the system is five.



For any selected component the original state before mixing has to be specified. If the component has been liquid, 'L' has to be selected (default). If the component has been in the vapor state, then 'V' should be activated.



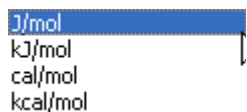
For *binary systems* it is possible to edit either data with a constant composition as environment condition or to specify a constant temperature. For ternary systems and higher only the latter may be specified.

If a *constant temperature* is given its value has to be entered. Furthermore for any data point the excess heat and

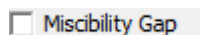
the composition (in mole fraction) has to be specified.

If a *constant composition* (mole fraction of the first component) is given, then for any data point the excess heat and a temperature has to be specified.

A *constant pressure* may be stated by selecting the corresponding checkbox and entering a valid pressure within the chosen pressure unit.



There are four possible *units* for the specified h^E data.



For isothermal data it is possible to state the existence of a *miscibility gap* under the given conditions (temperature and pressure, if given).

9. VE – Excess Volumes

Select the VE tab to edit excess volumes data (volume effects of mixing), densities and volumes of mixtures. The options and restrictions are nearly the same as for HE except the possibility to set the data type.

- Excess volume [cm^3/mol]
- Density [g/cm^3]
- Volume [cm^3/mol]

10. CPE – Excess Heat Capacities

To edit excess heat capacities and heat capacities of mixtures the CPE tab has to be selected. The options and restrictions are nearly the same as for HE except the possibility to set the datatype.

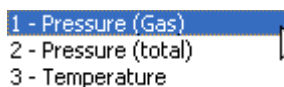
- Excess heat capacity [$\text{J}/\text{mol}\cdot\text{K}$]
- Heat capacity [$\text{J}/\text{mol}\cdot\text{K}$]

11. GLE – Gas Solubility Data

Gas solubility data can be edited by selecting the GLE tab.

This database doesn't use the temperature and pressure selection from the Units/Quality dialog because it stores the original units inside the database.

Currently one gas in up to three solvents may be specified. The first component will always be assumed to be the gas. So the maximum number of components in the system is four.



Constant Property: It is necessary to specify an environment condition. It is possible to choose between a partial pressure, a total pressure or a temperature value.

If a (partial or total) pressure has been selected as environment condition, then a temperature has to be specified for any data point. If a constant temperature has been specified, then it is possible to select partial pressure, total pressure or fugacity as table value.

Independent from being specified as constant or table value one of the various units has to be selected both for the pressure and for the temperature value(s).

The *Table Value* group can be used to customize the number and contents of the data table columns. Besides a temperature/pressure value up to two values may be specified.

1 - coeff.
3 - coef, ygas
4 - coef, xliquid
5 - coef, xliquid (gasfree)
7 - coef, wt%liquid
8 - coef, wt%liquid (gasfree)
9 - coef, kmol/m³

If there is one gas in one solvent given, then it is possible to specify the mole fraction of the gas in the vapor phase (selecting entry 3).

For two or more solvents the composition in the liquid phase must be specified. Choose entry 5 for mole fraction and entry 9 for kmol/m³ as unit. Usually the *real* mole fraction of all but the last solvent should be specified if either solubility coefficient no. 10 or no. 12 is selected (see below). Otherwise the *gas-free* mole fraction should be specified.

10 - X gas *10⁴
11 - mole gas/mole solvent
12 - X gas
13 - mole gas/mole solv.1
20 - Henry co. (P)
21 - Henry co. (f)
22 - Henry co. (P)/1000
24 - Pgas /Cgas [mol/dm³]
25 - Pgas /Cgas [mol/cm³]
30 - Ostwald co. [cm³/dm³]
31 - Ostwald co. [cm³/cm³]
40 - Kuenen co. [Ncm³/g]
41 - Kuenen co. [Ncm³/kg]
42 - Ncm³/g
50 - Bunsen co. [Ncm³/cm³]
51 - Bunsen [Ncm³/cm³]*100
52 - Bunsen co. [Ncm³/dm³]
60 - mole gas/g solvent
61 - mole gas/g solv.*10⁷
62 - mole gas/dm³ solvent
63 - mole gas/m³ solvent
64 - mole gas/1000 g solv.
70 - mg gas/kg solvent
80 - lamda [Ncm³/g/at]
90 - mole gas/mole amine

A solubility coefficient has to be specified. There are several different coefficient types that can be used. Please select the appropriate from the *Coefficient* list.

- 0 - unknown
- 1 - direct method
- 2 - pressure or volume difference
- 3 - comparative method
- 4 - gaschromatographic
- 5 - saturation method
- 6 - chemical (titration)
- 7 - capillary cell method
- 8 - static method
- 9 - microgasometric method
- 10 - piezometric method
- 11 - dilutor method
- 12 - phase equilibrium measurement
- 13 - other method

Finally a *Measurement Method* can be specified. If the method has not been published, then *unknown* should be selected. If the used method is not in the list, then *other method* should be chosen.

12. ELE – Vapor-Liquid Equilibrium Data of Electrolyte Systems

Select the ELE tab to edit vapor-liquid equilibrium data of electrolyte systems.

The ELE data bank allows to store information about systems containing up to four solvents and up to four salts. So the maximum number of components in the system is eight.

The field *Data Set Entries* describes how measured quantities may be combined to be processed. For ternary systems and higher (at least two solvents and at least one salt) the data types lower than 20 and in the thirties have to be used because they allow to specify a solvent composition. For binary systems (single solvent) the data types between 21 and 29 and higher than 40 must be used.

The abbreviation *xs* means the liquid solvent mole fraction (salt-free), *y* represents the vapor mole fraction, and again *T* is for temperature (in °C) and *P* is for pressure (in kPa). For binary systems a molal osmotic coefficient (*osm. coeff.*) or a mean molal activity coefficient (*gamma+-*) may be specified.

- 1 - <m> mole salt/1000g solvent
- 2 - <w%> weight% salt
- 3 - <Z> mole fraction salt
- 4 - <X3> mole fraction salt
- 5 - <M> mole salt/liter solvent

Concentration Scale: The abbreviation *conc* in the dataset entries list denotes the salt concentration. The selection list *Concentration Scale* can be used to select the desired unit. There are currently five different concentration units available. The units <m>, <w%> and <M> are self-explanatory. If the salt concentration is given in mole fraction without considering the charge of ions, then <Z> should be selected. The concentration unit <X3> is defined by the mole fraction of ions:

$$X_3 = \frac{n_3}{n_1 + n_2 + v n_3}$$

n: moles (n_1 , n_2 for solvents, n_3 for salt)

v: moles ions (referring to one mole completely dissociated salt – e.g. 2 for NaCl)

The value of the constant measured quantity (depending on the selected data type) has to be entered next to the *Concentration Scale* selector.

13. POW – Octanol-Water-Partition Coefficients

A data point in this database contains

- Temperature
- Logarithmic (decade) partition coefficient of a single component in the Octanol/Water system
- pH value of the liquid phase
- Experimental error (in log P_{ow} units)
- Measurement method (Normally unused, only predicted values should be marked)
- Quality code.

A dataset contains only single points.

14. EGLE – Gas Solubilities in Electrolyte Containing Systems

The EGLE database structure is the same as the GLE database. Salts and electrolytes are stored not with their special salt number but with their 'normal' component code.

Another slight difference is that reference links are not needed for this database since the normal literature numbers (LEAR) are used directly.

15. CRI – Critical Data of Mixtures

The CRI database contains temperatures, pressures, densities and compositions describing critical, near-critical or pseudo-critical information of mixtures.

The composition can be given in

1 MolePercent<100>
2 MoleFraction<1>
3 VolumePercent<100>
4 VolumeFraction<1>
5 MassPercent<100>
6 MassFraction<1>
7 Mole<>
8 kg<>
9 g<>

The numbers in angle brackets are used as conversion factors.

The concrete type of critical data has to be specified by the “State” selection box:

State
0 Not specified

The complete list of states:


State No.	Short Term	Description/Transition
0	Not specified	no specification on present/vanishing phases given
1	Liquid-vapor	Liquid-vapor to vapor/gas (VL > V/G)
2	UCEP	Upper critical end point
3	LCEP	Lower critical end point
4	Liquid-liquid	Liquid-liquid to liquid (LL > L)
5	to be defined	no specification on present/vanishing phases given
6	univariant (undef. phases)	Univariant line (present/vanishing phases not explicitly defined)

7	VLL	Vapor-liquid-liquid to two phases ($VLL > VL$, $VLL > LL$)
8	DCEP	Double critical end point
9	SVL TP	Solid-vapor-liquid (univariant) temperature/pressure data
10	VLL TP	Vapor-liquid-liquid (univariant) temperature/pressure data
11	Gas-gas	Gas-gas to gas ($GG > G$)
12	Near crit. TP (2PHS)	2-Phase temperature/pressure point near critical point
13	Univariant SLLV	Univariant point or data, solid-liquid-liquid-vapor phases
14	VL1L2L3	Vapor-liquid1-liquid2-liquid3 to three phases ($VLLL > VLL$, $VLLL > LLL$)
15	LLL > LL	Liquid-liquid-liquid to two phases ($LLL > LL$)
16	LLS TP	Solid-liquid-liquid univariant temperature/pressure data
17	S1S2LV TP	Solid-solid-liquid-vapor univariant temperature/pressure data
18	CST (LLE)	Critical solution temperature ($LL > L$)
19	LCST (LLE)	Lower critical solution temperature ($LL > L$)
20	UCST (LLE)	Upper critical solution temperature ($LL > L$)
21	SL1G	Solid-Liquid1-Gas univariant temperature/pressure data
22	SL2G	Solid-Liquid2-Gas univariant temperature/pressure data
23	LL > L	Transition from two to one liquid
24	HL1L2	Hydrate-liquid-liquid (univariant) temperature/pressure data
25	HL1G	Hydrate-liquid-vapor/gas (univariant) temperature/pressure data
26	Quadruple point	Four phases present under given (univariant) conditions
27	Structural transition point	Hydrate I to Hydrate II transition
28	L1L2S2 TP	Liquid-liquid-solid-temperature/pressure data
29	K-point ($L1-L2=V$)	Liquid-liquid to liquid-vapor ($L1L2 > L1V$)
30	HL1L2V	Hydrate-liquid-liquid-vapor (univariant) temperature/pressure data
31	HL2G	Hydrate-liquid-vapor/gas (univariant) temperature/pressure data
32	HiceG	Hydrate, Ice, Gas)

The “Table Entries” selection drop-down-boxes allow to define up 7 columns. Beside the main data types it is possible to add errors for them.

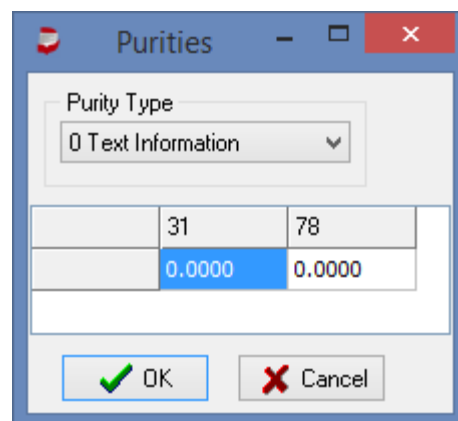
For every table entry it is necessary to define the unit.

Table Entries	Units
1 T Temperature	1 K
51 ET Error in Temperature	1 K
2 P Pressure	1 kPa
52 EP Error in Pressure	1 kPa
3 V Volume	1 cm ³ /mol
53 EV Error in Volume	1 cm ³ /mol
0 Not available.	

The button  opens a dialog where the purities of the used pure component properties can be specified.

The “Purity Type” selection menu contains the following entries:

- 0 Not specified
- 1 Weight Percent
- 2 Volume Percent
- 3 Mol Percent
- 4 Area Percent
- 5 Refractive Index
- 6 Percent



The Purities dialog box shows a 'Purity Type' dropdown menu set to '0 Text Information'. Below it is a table with two columns and two rows. The first row contains the values '31' and '78'. The second row contains '0.0000' and '0.0000'. At the bottom are 'OK' and 'Cancel' buttons.

Purity Type	
31	78
0.0000	0.0000

Figure 21 Purities

Additional entries are a measurement method and an unspecified comment.

The data are stored in original units and in original notation.

16. ESLE – Salt Solubilities

The ESLE database contains information on the solubility of salts and electrolytes in pure components and mixtures.

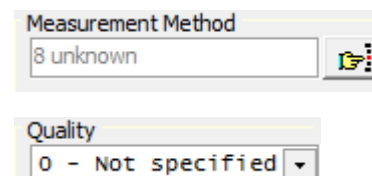
The common component/salt (see above) list has to be separated in solved and precipitated components and solved and precipitated salts.

Components			Salts		
174	<input checked="" type="checkbox"/> Solved	<input type="checkbox"/> Precipitated	1000052	<input checked="" type="checkbox"/> Solved	<input checked="" type="checkbox"/> Precipitated
			1000104	<input checked="" type="checkbox"/> Solved	<input checked="" type="checkbox"/> Precipitated
			1000115	<input checked="" type="checkbox"/> Solved	<input type="checkbox"/> Precipitated

Figure 22 Solved/Precipitated Classification

The salts are shown with their salt number increased by 1 million to make them easily identifiable.

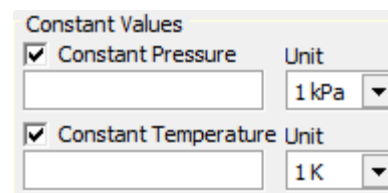
A ESLE dataset also contains information in the measurement method and the quality of the experimental data.



The Measurement Method field contains the text '8 unknown'. The Quality dropdown menu is set to '0 - Not specified'.


It is possible to define a constant temperature and a constant pressure. For both properties it is necessary to define the unit.


Pressure and temperature can either be constant or in the data table when they change for every data point.




The Constant Values dialog box has two sections. The first section is for 'Constant Pressure' with a checked checkbox and a unit dropdown set to '1 kPa'. The second section is for 'Constant Temperature' with a checked checkbox and a unit dropdown set to '1 K'.

The data table can be build from the selection box “New Table Entry”.

 **Add** Adds a new column

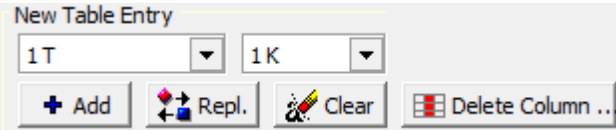
 **Repl.** Changes the currently selected column

 **Clear** Clears the entire grid

 **Delete Column ..**

Deletes a column specified in a pop-up menu:

1 T [°F]
2 SOLUB [g/100gSolv]



The dialog box titled "New Table Entry" contains two dropdown menus. The first dropdown menu shows "1 T" and the second shows "1 K". Below the dropdowns are four buttons: "Add" (with a plus icon), "Repl." (with a double-headed arrow icon), "Clear" (with an eraser icon), and "Delete Column .." (with a red X icon).

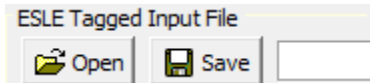
Possible table entries are

- Temperature
- Pressure
- Salt solubility (SOLUB) in molar and mass based units
- Solvent composition (COMPOS)
- Component and salt number of precipitated component or salt (SALT_PREC, COMP_PREC)
- Time needed for equilibrium (EQTIME)
- Liquid density (LIQDEN)
- Salt solubility (SOLUBV) in volume based units
- Component solubility (SOLUBC)

1 T
2 P
3 SOLUB
4 COMPOS
5 SALT_PREC
6 COMP_PREC
7 EQTIME
8 LIQDEN
9 SOLUBV
10 SOLUBC

Experimental data are stored in original units.

A special feature is the possibility to read and save a tagged file format. These files must have the extension “.sle” and are pure text files.



The dialog box titled "ESLE Tagged Input File" contains two buttons: "Open" (with a folder icon) and "Save" (with a floppy disk icon). There is also an empty text input field to the right of the buttons.

17. AAE – Adsorbent/Adsorptive Equilibria

The AAE database contains data on standard components adsorbed on zeolites, coals, molecular sieves etc.

The database content is specified by data type number.

Datatype	Description
Pure Adsorptives	

Datatype	Description
11	isothermal data (temperature = const.)
12	isostere data (amount = const.)
13	isobaric data (pressure = const.)
Binary Adsorptives	
21	isobaric data (pressure, temperature = const.)
22	isothermal data (vapor phase mole fraction, temperature = const.)
23	isothermal data (relative saturation with water, temperature = const.)
24	isothermal data (volume ratio at the beginning, temperature = const.)
25	isothermal data (pressure is nearly const., temperature = const.)
Ternary Adsorptives	
31	variation of the total pressure and fixed vapor composition (Y_i , temp. = const.)
32	const. total pres. and const. vapor composition ratio (pres., temp., one Y_i = const.)
33	const. total pres. and const. vapor composition ratio (pres., temp., all Y_i = const.)
34	const. total pres. (pressure, temp. = const)
35	variation of total pressure at constant partial pressure of water as component (temp., Y_{H_2O} = const.)
36	const. volume ratio of adsorptives at the beginning of the measurement (temp., beginning volume ratio = const.)
37	only constant temperature (temperature = const.)

These datatypes determine the content of the data table.

Adsorbent Details

15 

The adsorbent is defined by two entries. One number is a coarse specification of the adsorbent type, zeolite x, activated carbon etc. This number is selected in the common components/salts/adsorbents list of EditMixtureData. The other number references an entry in a adsorbent details list where quantitative data for the special adsorbent is specified.

This adsorbent details list contains:

1. Adsorbent number (class)
2. Unit cell molecular weight – if available
3. Reference number/s
4. Four fields with details. These fields are pure text, but contain some tags listed in the “Short Terms” box. For some historic reasons empty fields are filled with the string “E-M-P-T-Y”.
5. Two text fields reserved for kinetic information – which are currently unused.

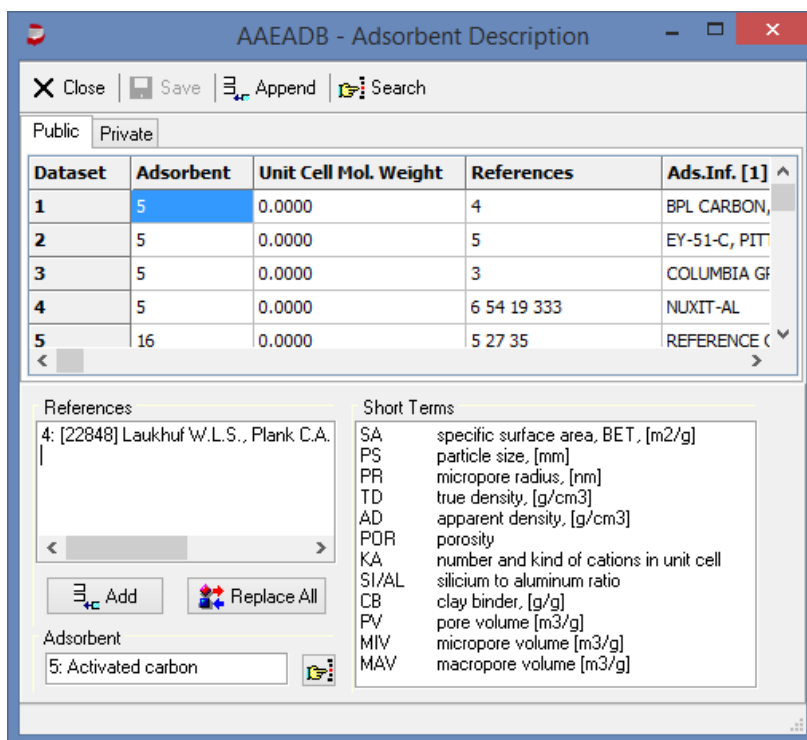
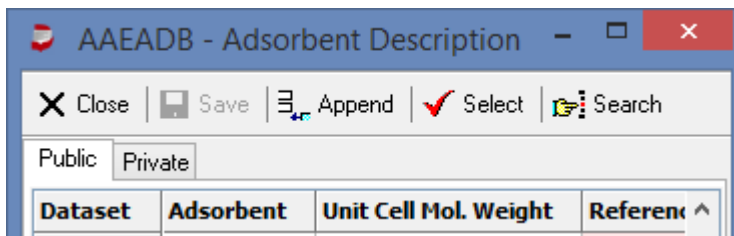
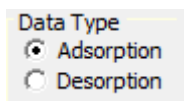


Figure 23 Adsorbent Details

An item can be selected by selecting the “Select” button or by double-clicking the appropriate line.

The database contains information on adsorption and desorption. The process type has to be specified in the “Data Type” radio box.



The “Original Unit” selection boxes specify the units of pressure, temperature, and amount in the original publication. The database itself only stores temperatures in [K], pressures in [kPa], and adsorbed amounts in [mmol/g].

18. Polymers

The polymer data bank contains

- Vapor-liquid equilibria
- Liquid-liquid equilibria (mixing gaps)
- Solubilities
- Excess and dilution enthalpies
- Densities and volumes
- Swelling
- Partition coefficients

- and few other data

for polymers and polymer containing systems.

Property	Phase	Compound(s)	Unit	Value	Remove?
1 Temperature	0 System	0 All	37 K	393.15	Remove

Property	3 [Concentration] ...	31 [Activity] ...
Phase	1 [Liquid] ...	0 [System] ...
Compound	1 [Polymer B] ...	2 [Component A] ...
Unit	2 [weight fraction] ...	73 [Partial Pressure (I)/Saturation Vapor ...]
1	0.6021	0.7425
2	0.6418	0.6914
3	0.6673	0.6777
4	0.7021	0.6202

Figure 24: Polymer Data Set Editor

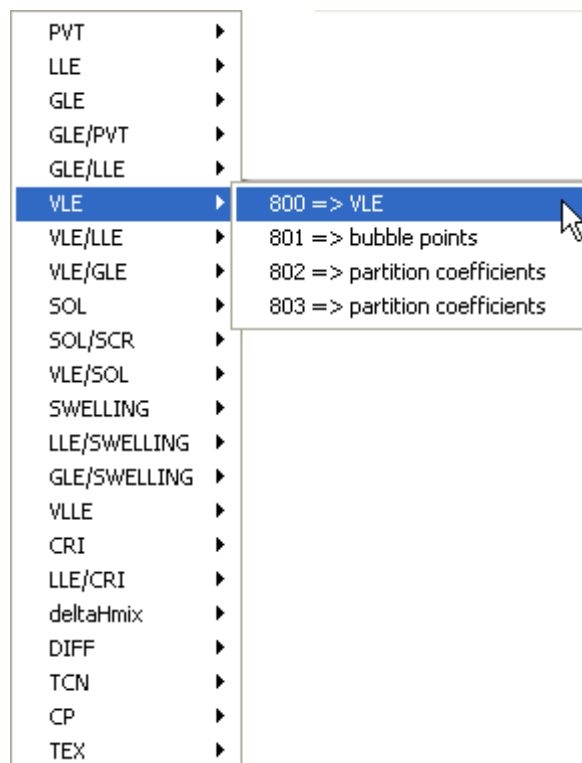
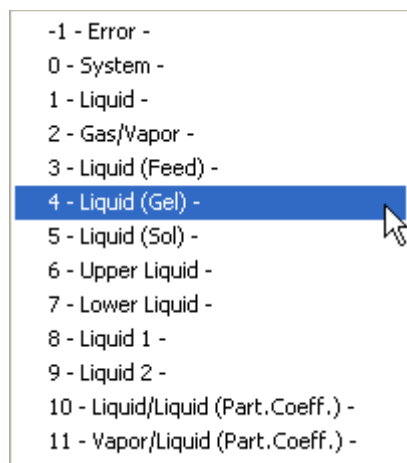
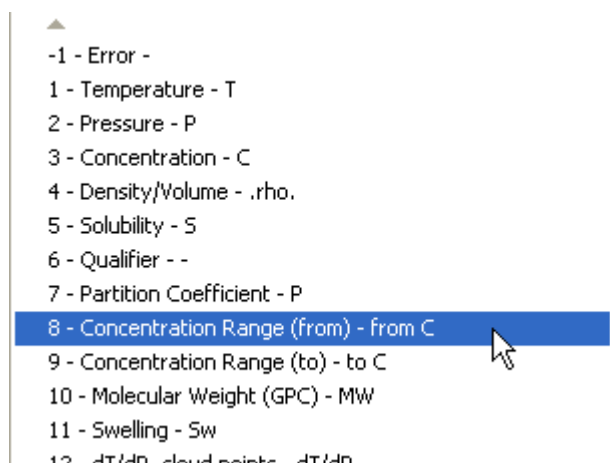
The “Data Type” is a number specifying the properties of the data set. The edit field has a context menu where available data types are listed.

This data type does describes the content of the data set but the table items can be selected independent of this number. This is a possible pitfall since table items might be specified which have no meaning for the specified data type.

Constants

A polymer data set might contain as many constants as needed.

A constant is defined by its property, phase, related components, a unit, and its value. All data have to be entered via context menus.



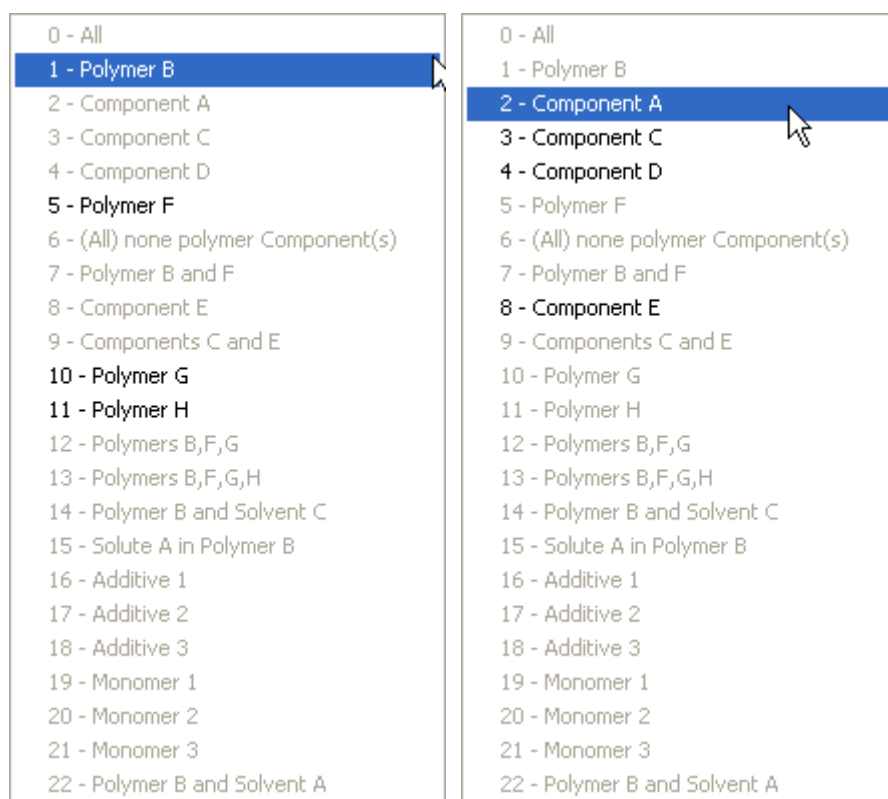
0 - All	1 [mole fraction]
1 - Polymer B	2 [weight fraction]
2 - Component A	3 [volume fraction]
3 - Component C	4 [ccm/g Polymer]
4 - Component D	5 [g/g amorph Polymer]
5 - Polymer F	6 [g(A)/100g Polymer]
6 - (All) none polymer Component(s)	7 [weight fraction, gasfree]
7 - Polymer B and F	10 [g Gas / kg Polymer]
8 - Component E	11 [g/100ccm]
9 - Components C and E	12 [Nccm/g Polymer]
10 - Polymer G	14 [mol% (free of polymer)]
11 - Polymer H	16 [mole/1000ccm]
12 - Polymers B,F,G	
13 - Polymers B,F,G,H	
14 - Polymer B and Solvent C	
15 - Solute A in Polymer B	
16 - Additive 1	
17 - Additive 2	
18 - Additive 3	
19 - Monomer 1	
20 - Monomer 2	
21 - Monomer 3	
22 - Polymer B and Solvent A	

The component description is not a direct relation to a specific components but rather a sort of index. This index (“Polymer B”, “Additive A” etc.) are specified in the “Polymer/Component Details” page.

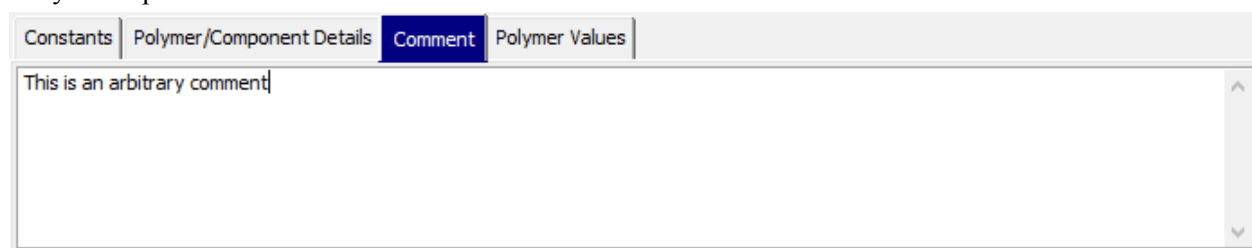
Constants	Polymer/Component Details	Comment	Polymer Values						
#Polymer/Comp't	#Index	WMN	WMW	WMZ	WMETA	Polydispersity Index	Additional Info.	Additional Info.	Trade Name
3000004	1 Polymer B	1710	3710			2.16959064327485			polyethene
27	2 Component A								

This table also allows to enter several details for the polymers like mean molecular weights and a polydispersity index. Two fields are reserved for any additional information and one field is for a trade name.

The “#Index” column is used for specifying a component or polymer specific index and the entries are selected via context menus.



Only valid entries are selectable. The “Comment” page is a place for entering arbitrary text which does not fit into any other place.



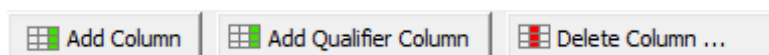
The “Polymer Values” is twin of the “Constants” page but intended for polymer specific values only.

Constants	Polymer/Component Details	Comment	Polymer Values	
#Index	Property	Unit	Value	Remove?
1 Polymer B	30 Reaction Rate	57 1/s	658	<div>Remove</div>

New Value

Specifying the Columns of the Table

The buttons



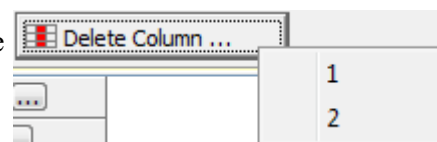
allow to add and remove columns. The “Add Column” function adds always the same column type:

These four entries represent the property, the phase, the compound or compounds, and the unit of the property. The ellipsis buttons allow to change the values by context menus.

3 [Concentration] ...
1 [Liquid] ...
1 [Polymer B] ...
2 [weight fraction] ...

The adds a special column containing only unspecified values, text or numbers. Only a single qualifier column can be added.

The button displays a context menu with the column numbers.



Exploring the Definition Lists

The polymer data bank uses in several cases indexes instead of a complete and repeated storage of some data.

Polymer Components		Polymer Datatypes	Polymer Phases	Polymer Properties	Polymer Units
Index	Component Specification				
0	All				
1	Polymer B				
2	Component A				
3	Component C				
4	Component D				
5	Polymer F				

The index numbers are pointing to entries in the definition lists for

- Component Specification (index code)
- Data Types (like VLE, LLE, SLE, etc.)
- Phases (liquid, vapor, solid, etc.)
- Properties (Temperature, Concentration, etc.)
- Units (specific lists for all properties)

These lists are currently not extendable.